

Exam I Study Guide

- **Physical Changes vs. Chemical Changes (Can you undo it?)**
 - Physical changes result in changes of appearance, but not composition.
 - Melting, evaporation, freezing, etc.
 - Chemical changes result in changes of composition of matter. (Chemical reaction)
 - Burning, mixing, dissolving, etc.
- **Physical Properties vs. Chemical Properties**
 - Physical properties can be described by observing the matter without a change in the composition.
 - Color, odor, density, temperature, etc.
 - Chemical properties can only be detected by altering the composition.
 - Flammability, oxidization, etc.
- **Atoms vs. Molecules**
 - Atoms are the smallest unit of a given element that carry the properties of that element.
 - Molecules are composed of two or more atoms.
- **Elements vs. Compounds**
 - Elements occur in nature and cannot be broken down to simpler particles through ordinary means.
 - Compounds are the combination of two or more elements.
- **Accuracy vs. Precision**
 - Accuracy refers to the agreement of a measurement with a known or accepted value. (Bullseye)
 - Correctness
 - Precision refers to the agreement of a measurement with other measurements in a given set of measurements. (Missing the bullseye but landing all darts in one place)
 - Repeatability or reproducibility
- **Rules Concerning Significant Figures**
 - All nonzero digits ARE significant.
 - Zeros between nonzero digits ARE significant.
 - Zeros at the beginning of a number are NOT significant; they only indicate the position of the decimal point.
 - Zeros at the end of a number to the right of a decimal point ARE significant.
 - Zeros at the end of a number with no decimal point are ASSUMED to be NOT significant, UNLESS we indicate their significance by placing a line over the last zero that is significant.

General Chemistry 1 Final Exam Study Guide

S Ben Porath



General Chemistry 1 Final Exam Study Guide:

The Enigmatic Realm of **General Chemistry 1 Final Exam Study Guide**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing lacking extraordinary. Within the captivating pages of **General Chemistry 1 Final Exam Study Guide** a literary masterpiece penned with a renowned author, readers embark on a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting impact on the hearts and minds of those who partake in its reading experience.

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