

Molarity can be calculated from grams of solute by converting to moles of solute.

$$\text{molarity} = \frac{\text{number of moles of solute}}{\text{liter of solution}} = \frac{\text{moles}}{\text{liter}}$$

$$\text{moles solute} = \frac{\text{g solute}}{\text{molar mass}}$$

$$\text{Molar mass} = \frac{\text{g}}{\text{mole}}$$

$$\text{Molarity} = \frac{(\text{g solute}) \left( \frac{1 \text{ mole}}{\text{molar mass g}} \right)}{\text{Liter of solution}} = \frac{\text{moles}}{\text{liter}}$$

$$\text{molarity} = \frac{\text{g solute}}{\text{molar mass} \times \text{liter of solution}} = \frac{\text{moles}}{\text{liter}}$$

# Grams And Liters Molarity Calculations Answers

**Michael L. Bishop**



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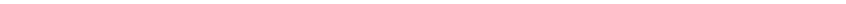
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