

HP JetDirect 170X External Print Server and HP JetAdmin Software

Part of HP's industry-leading network printing solutions



Connect any HP or non-HP printer with a parallel port* anywhere on your network.



HP network printing solutions reduce the total cost of ownership and improve workflow.



HP JetAdmin

Included with the purchase of any HP JetDirect product.

HP JetDirect 170X External Print Server

for Ethernet Networks

A simple, affordable solution for sharing printers on a network

The HP JetDirect 170X External Print Server is ideal for a small business or office that wants to connect any printer with a parallel port, or a supported HP All-in-One peripheral*, to its growing network, simply and affordably. Advantages include faster document printing and the flexibility to locate printers anywhere on the network.

HP JetAdmin Software

A simple tool for efficiently installing network printers

HP JetAdmin provides remote installation of HP JetDirect-connected printers within Microsoft® Windows® 95, Windows 98 (when released), Windows NT, and Novell® NetWare® environments. These capabilities save network managers time, increase user productivity, and reduce the overall cost of ownership.

HP JetDirect 170X External Print Server for Ethernet Networks

New for Windows NT 4.0

The HP JetDirect 170X External Print Server supports IP multicasting to allow **Simple Device Installation**. Working with an Internet Protocol and your DHCP server, HP Network Printer Installer for Windows NT software automatically discovers and installs new printers. Plug in your printer and it works!

Best Value

- Industry's lowest price-to-performance print servers
- The most cost-effective way to share network printers
- Reduced waiting for print jobs because files print up to six times faster than via a file server or PC connection

Compatible

- Support for Microsoft Windows 95, Windows 98 (when released), Windows NT, Novell NetWare, IntraNetWare, and Artisoft LANtastic environments
- Ideal for connecting any printer with a parallel port*, such as an HP LaserJet, HP Pro-Professional Series, or HP DeskJet printer
- Fully compliant with Novell Directory Services (NDS), plus support for Novell Distributed Print Services (NDPS)
- Network scanning function for supported HP All-in-One peripherals†
- Embedded Web server for status and configuration using a Web browser
- Optimized for small business

Simple

- Quick-install guide for easy setting up
- Simple installation and remote printer configuration via HP JetAdmin software
- Microsoft Windows NT Simple Device Installation via HP Network Printer Installer for Windows NT
- Convenience of placing shared network printer or HP All-in-One peripheral close to users



Supported Operating Systems

- Microsoft Windows 95
- Microsoft Windows 98 (when released)
- Microsoft Windows NT v3.51, 4.0
- Novell NetWare v3.x, 4.x
- Novell IntraNetWare
- Artisoft LANtastic v7.0

Product Name and Number



HP J0258A

The HP JetDirect 170X External Print Server for Ethernet networks includes an RJ-45 port for 10Base-T unshielded twisted-pair cable.

Warranty

One-year, return-to-HP warranty

Product package includes power supply, quick-install guide, and HP JetAdmin software on CD-ROM.

*Except 500 printers or HP DeskJet 720, 820, and 950 printers.
†HP LaserJet 7100A.

Hp Jetdirect 170x Office Connect Manual

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web the formula for molality and molality units molality is measured in milligrammes m or milligrammes per kilogramme mol kg the molality equation is a mathematical formula that describes how much a substance weighs in this equation m moles solute per kilogramme of solution

342g of 20% by mass of Ba(OH)_2 solution specific gravity 1.057 is reacted with 1200 ml of 2M HNO_3 . If the final density of solution is same as pure water, then molarity of the iron in resulting solution which decides the nature of the above solution is

- 1.025M
- 0.5M
- 0.888M
- none of

web what is the formula for converting molarity to molality molality is equal to the product of molarity divided by the density of the solution at that temperature $m = \frac{M}{d}$ which is more important molality or molarity both molality and molarity are concentration units and they are used interchangeably

web correct answer c both a and b q2 the density of 1 m solution of hcl is 1 0585 g ml the molality of the solution is 1 0585 1 0 10 0 0585 correct answer b 1 explanation molality begin array l m left frac 1000 m 1000d mm right end array

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web molarity is the number of moles of a substance per litre of solution also known as molar concentration a capital m signifies solutions labelled with molar concentration a 1.0 M solution contains 1 mole of solute per litre of solution molality is the number of solvent moles per kilogram

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web 1 molarity $m = \frac{\text{mass}}{\text{volume}}$ mass molar mass $\times 0.4500 \text{ L} = 0.825 \text{ g}$ 141.9579 g/mol $\times 0.0129 \text{ m}^3 = 0.825 \text{ g}$ 141.9579 g/mol 0.00581158 mol 0.00581158 mol 0.4500 kg 0.0129 m³ mole fraction $n_2 = \frac{0.825 \text{ g}}{141.9579 \text{ g/mol}} = 0.00581158 \text{ mol}$ $n_1 = \frac{450.0 \text{ g}}{18.015 \text{ g/mol}} = 24.97918401 \text{ mol}$

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web answer the correct option is c explanation the molarity of a mixture m_{mix} can be calculated using the following formula $m_{\text{mix}} = \frac{m_1 + m_2 + m_3}{V_1 + V_2 + V_3}$ $1.5 \times 480 + 1.2 \times 520 + 0.8 \times 720 = 1000$ 1.344 M q5 with increase in temperature which of these changes a molality b fraction of solute present in water

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web determine the molarity for each of the following solutions 0 444 mol of CoCl_2 in 0 654 l of solution 98 0 g of phosphoric acid H_3PO_4 in 1 00 l of solution 0 2074 g of calcium hydroxide Ca(OH)_2 in 40 00 ml of solution 10 5 kg

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