

Chapter 2 Practice Problems

1. What does "X" represent in the following symbol?



- (A) silicon
B) sulfur
C) zinc
D) ruthenium
E) nickel

2. Determine the number of protons, neutrons and electrons in the following:



3. Determine the number of protons, neutrons and electrons in the following:



4. What element is defined by the following information?



5. What species is represented by the following information?



6. Predict the charge that an aluminum ion would have.



7. Predict the charge that a calcium ion would have.



8. Calculate the atomic mass of silver if silver has 2 naturally occurring isotopes with the following masses and natural abundances:

Ag-107	106.90509 amu (51.8%)	51.84%	= 55.42
Ag-109	108.90476 amu (48.1%)	48.16%	= 52.78
			> = 108.2

9. There are 54 electrons, 54 protons, and 78 neutrons in an atom of ${}_{54}^{132}\text{Xe}$.

Gradpoint Test Answers Chemistry

Ensheng Dong

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Gradpoint Test Answers Chemistry Book Review: Unveiling the Power of Words

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