

Chemistry Chapter 5 Exam

Multiple Choice

1. Which of the following is not a transition metal?

- _____ a. Potassium
- _____ b. Lanthanum
- _____ c. Cadmium
- _____ d. Manganese
- _____ e. Copper
- _____ f. Silver
- _____ g. Transition Metals
- _____ h. Actinides
- _____ i. Alkali Metals
- _____ j. None of the above

- _____ k. The elements with atomic numbers 49-52, 74
- _____ l. The p-block elements together with the s-block elements, Groups 1-2, 13-18
- _____ m. One half of the elements between the noble gases, elements that are located together
- _____ n. Properties that relate to the element of an ion
- _____ o. The fact that atoms that are adjacent, physical and chemical properties, the elements change periodically with increasing numbers of the elements
- _____ p. The elements with atomic numbers 57-71, 89-103
- _____ q. The s-block elements are metals with typical metallic properties
- _____ r. Transition ion
- _____ s. Periodic law
- _____ t. The elements of Groups 1-10

Multiple Choice

2. Which type of element is not a metal?

- _____ a. alkali earth metals
- _____ b. alkali metals
- _____ c. transition metals
- _____ d. transition metals

3. Which group of elements is not a metal?

- _____ a. Halogens
- _____ b. Noble Gases
- _____ c. Lanthanides
- _____ d. Actinides

4. Which is not a metal, or the other of the periodic table?

- _____ a. Noble Gases

Modern Chemistry Chapter 5 Test Answers

Ralph H. Petrucci



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