

EXPERIMENT NO: SINGLE STAGE LEACHING

Aim: To determine the efficiency of single stage leaching experiment for leaching of NaOH and CaCO_3 .

Apparatus: Glass beakers, glass rod, measuring cylinder and stirrer.

Chemicals: NaOH, CaCO_3 , Phenolphthalein and water.

Procedure:

- Clean all apparatus with water.
- Prepare the mixture for extraction by mixing 5 gm NaOH and 3.5 gm CaCO_3 and transfer it to a beaker.
- Add 200 ml of water into the beaker and stir the solution for about 10 min.
- Allow the mixture to settle and a clear liquid layer will form at the top, this is the extract layer.
- Take 25 ml of the clear solution and titrate it against 0.5 N HCl with phenolphthalein indicator.
- Note down the observation and do the calculation.

Observation:

- | | |
|-----------------------------------|---|
| 1. Weight of feed NaOH | - |
| 2. Weight of feed CaCO_3 | - |
| 3. Weight of solvent (water) | - |
| 4. Volume of clear solution | - |
| 5. Indicator | - |
| 6. Volume of extract | - |
| 7. Titrate reading | - |
| 8. Normality of HCl solution | - |

Calculation:

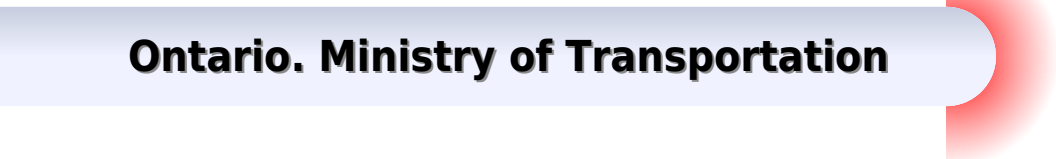
i. For clear solution normality, $N_1 = \frac{N_2 V_2}{V_1}$

ii. Amount of NaOH in clear solution (extract)

$$= N_1 \times \text{volume} \times \text{gm equivalent of NaOH}$$

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