

# **FORMULAS WITH POLYATOMIC IONS**

Name \_\_\_\_\_

Matching the horizontal and vertical axes, write the formulas of the compounds with the following combination of ions. The first one is done for you.

|                  | $\text{OH}^-$                            | $\text{NO}_3^-$ | $\text{CO}_3^{2-}$      | $\text{SO}_4^{2-}$      | $\text{PO}_4^{3-}$      |
|------------------|------------------------------------------|-----------------|-------------------------|-------------------------|-------------------------|
| $\text{H}^+$     | $\text{HOH}$<br>( $\text{H}_2\text{O}$ ) | $\text{HNO}_3$  | $\text{H}_2\text{CO}_3$ | $\text{H}_2\text{SO}_4$ | $\text{H}_3\text{PO}_4$ |
| $\text{Na}^+$    |                                          |                 |                         |                         |                         |
| $\text{Mg}^{+2}$ |                                          |                 |                         |                         |                         |
| $\text{NH}_4^+$  |                                          |                 |                         |                         |                         |
| $\text{Ca}^{+2}$ |                                          |                 |                         |                         |                         |
| $\text{K}^+$     |                                          |                 |                         |                         |                         |
| $\text{Al}^{+3}$ |                                          |                 |                         |                         |                         |
| $\text{Pb}^{+4}$ |                                          |                 |                         |                         |                         |

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**CL Gary**



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