

Introduction to Carbon Bonding

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Carbon Bonding

Chemistry fundamentals

Writing and using chemical formulae is an essential skill in chemistry. It is a language that allows chemists to communicate their ideas and findings. In this section, we will learn the basics of writing and using chemical formulae.

Electron configurations and valence bonding theory

One of the most important concepts in chemistry is the arrangement of electrons in an atom. This is described by the electron configuration. The outermost electrons, known as valence electrons, are responsible for the chemical properties of an element.

Orbital structures

Orbitals are regions of space where there is a high probability of finding an electron. They are described by mathematical functions. The shape and size of an orbital depend on the energy level and the angular momentum of the electron.

Molecular orbitals and molecular geometry

Molecular orbitals are formed by the combination of atomic orbitals. They describe the distribution of electrons in a molecule. The geometry of a molecule is determined by the arrangement of its atoms in space.

Carbon bonds in organic compounds

Carbon is a unique element because it can form four covalent bonds. This allows it to form a wide variety of compounds, including the organic molecules that are essential for life.



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