

Periodic Table Basics

Step 1: Complete the square for each element by filling in the atomic number, name, & atomic mass.

Step 2: Determine the number of protons, neutrons, and electrons in an atom of each element.

Step 3: Identify if the element is a solid, liquid, or gas at room temperature.

Step 4: Give the melting (M.P.) and boiling points (B.P.) in degrees Celsius.

Step 5: List at least three physical or chemical properties for each element.

Step 6: List at least three uses for each element.

Step 7: Draw a Bohr diagram and Lewis Structure to show the arrangement of electrons and the number of valence electrons.

	P = N = E =	O = L = G =
	M.P. = B.P. =	
Properties		
Uses		
Bohr Diagram Lewis Structure		

Step 8: Use the following colors to shade in the square for each element. You should ONLY color the small square in the upper left-hand corner and not the entire card.

Green = Li & Na
Orange = B & Al

Pink = O & S
Red = C & Si

Blue = Be & Mg
Tan = N & P

Purple = F & Cl
Yellow = He, Ne, & Ar

Step 9: Cut the cards apart and arrange according to atomic number in the pattern shown. Once you have the cards arranged in the correct order, glue them to a large sheet of construction paper.

1							2
3	4	5	6	7	8	9	10
11	12	13	14	15	16	17	18

Done? Answer the questions on the worksheet using the information on your Periodic Table!

Heat Calculations Answers Physical Science If8767

S. M. Niven



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Analysis of Heat Equations on Domains. (LMS-31) El-Maati Ouhabaz, 2004-10-31 This is the first comprehensive reference published on heat equations associated with non self adjoint uniformly elliptic operators The author provides introductory materials for those unfamiliar with the underlying mathematics and background needed to understand the properties of heat equations He then treats L_p properties of solutions to a wide class of heat equations that have been developed over the last fifteen years These primarily concern the interplay of heat equations in functional analysis spectral theory and mathematical physics This book addresses new developments and applications of Gaussian upper bounds to spectral theory In particular it shows how such bounds can be used in order to prove L_p estimates for heat Schrödinger and wave type equations A significant part of the results have been proved during the last decade The book will appeal to researchers in applied mathematics and functional analysis and to graduate students who require an introductory text to sesquilinear form techniques semigroups generated by second order elliptic operators in divergence form heat kernel bounds and their applications It will also be of value to mathematical physicists The author supplies readers with several references for the few standard results that are stated without proofs **The Theory of Heat** Thomas Preston, 1919 **The Theory of Heat** Thomas Preston, 1894 Calculations in Heat, Light and Sound ... S. M. Niven, 1964 **Teaching Heat Transfer and Heat Exchange** Brian Melvin Evenson, 2004 Analysis of Heat Equations on Domains El Maati Ouhabaz, 2005 **Heat and Thermodynamics** Mark Waldo Zemansky, Richard H. Dittman, 1997 This respected text deals with large scale easily known thermal phenomena and then proceeds to small scale less accessible phenomena The wide range of mathematics used in Dittman and Zemansky's text simultaneously challenges students who have completed a course in impartial differential calculus without alienating those students who have only taken a calculus based general physics course Examples of calculations are presented shortly after important formulas are derived Students see the solutions of problems related to the formulas Actual thermodynamic experiments are explained in detail The student sees the applicability of abstract thermodynamic concepts and formulas to real situations The New Heat Theorem Walther Nernst, 1969 **Worked Problems in Heat, Thermodynamics, and Kinetic Theory for Physics Students** L. Pincherle, 1966 Worked Problems in Heat Thermodynamics and Kinetic Theory for Physics Students is a complementary to textbooks in physics This book is a collection of exercise problems that have been part of tutorial classes in heat and thermodynamics at the University of London This collection of exercise problems with answers that are fully worked out deals with various topics This book poses problems covering the definition of temperature such as calculating the assigned value of the temperature of boiling water under specific conditions This text also gives example of problems dealing with the fir

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