

Chemical Formulas

Take the chemical formula number of atoms of each element in a chemical formula

Example: 1 Carbon atom (reading)

Formula represents the number of atoms of each element in a compound

Example: 2 Hydrogen atoms (reading)

Formula represents the number of atoms of each element in a compound

Atomic Mass

The atomic mass of an atom is approximately equal to the sum of the protons and neutrons

Reading

- 1 atom of each element
- 1 atom of each element
- 1 atom

Atomic Mass

- 1 atom of each element
- 1 atom of each element
- 1 atom

Gbp Note Taking Guide Episode 601 Answers

Robert R. Redfield



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