

#### GRADE 9 NATURAL SCIENCES JUNE TEST

SECT						
- 6	0.000	s options are provided as answers rect answer and write CNLY the L ION NUMBER (1.1.1-1.1.10) on y	ETTER	(A-D) next to the		
1,1,1	Which of the following can you NOT learn from a chemical formula?					
	A	Elements in the compound	8	Phase of the element		
	C	Ratio of atoms	D	Total number of atoms		
1.1.2	Which of the following is NOT a sub-atomic particle?					
	A	Electrons	. 8	Protons		
	0	Neutrons	0	Nucleons		
1.13	What is the prefix for 6 stores?					
	A	Hepta-	8	Hexa-		
	C	Pero-	0	Tri-		
1.1.4	An element has an atomic number of 30 and mass number of 65. How many nucleons will if have?					
	A	30	8	35		
	C	65	0	95		
1.1.5	Element at group 15, period 2 on the periodic table					
	A	Carbon	8	Nitrogen		
	100	Silicon	. 0	Oxygen		

	180	Silicon		Lithium	
	C	Magnesium	0	Carbon	
17	Subs	stances on the left side of a chemi	cal equa	flort:	
	A	Products		Reactants	
	C.	Metals	0	Non-metals	
33	Which	it of the following is NOT a way to	preven	t rust?	
	A	Galvanisation		Electrolysis	
	¢	Paint	0	Ollor grease	
19	Sut	stance with a pH lower than 7 wil	ibe dan	sified as:	
	A	Lemon Juloe		Neutral	
	¢	Apde	0	Akaine	
3.30	An	neutral substance would turn	with	universal indicator.	
	A	Green		Volet.	
	ė.	Colourless		Red	[Mat]
		e the correct TERM for the follow			
		e the correct TERM for the follow next to the QUESTION NUMBER			
T	ERM		b(1.2.1-	(2.7) on your answer	
	ERM Mea	need to the QUESTION NUMBER	b(1.2.1-	(2.7) on your answer	
21 22	ERM Mea Reac	need to the QUESTION NUMBER sure of how strongly acidic or also	t (1.2.1- line a su	(2.7) on your answer	
21 22 22 23	ERM Mea Rea Nam	next to the QUESTION NUMBER sure of how strongly acidic or alka stion of a substance with oxygen.	t (1.2.1- line a su lie.	(27) on your answer	
21 22 23 24	ERM Mea Read Nam Proc	next to the QUESTION NUMBER sure of how strongly acidic or also stion of a substance with oxygen. e for group 18 on the periodic tab	t (1.2.1- line a su lie.	(27) on your answer	
2! 22 23 24 25	ERM Mea Reac Nam Proo H <sub>2</sub> (	next to the QUESTION NUMBER sure of how strongly acidic or alka stion of a substance with oxygen. e for group 18 on the periodic tab ess whereby compounds can be i	t (1.2.1-) Vine a su Vie Inchen d	(2.7) on your answer to be a second of the s	

1.1.6 The following is an example of a semi-metal:

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**G Thomas** 

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## **Table of Contents Natural Science Grade 9 External Exam Papers**

- 1. Understanding the eBook Natural Science Grade 9 External Exam Papers
  - The Rise of Digital Reading Natural Science Grade 9 External Exam Papers
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Natural Science Grade 9 External Exam Papers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Natural Science Grade 9 External Exam Papers
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Natural Science Grade 9 External Exam Papers
  - Personalized Recommendations
  - $\circ\,$  Natural Science Grade 9 External Exam Papers User Reviews and Ratings
  - Natural Science Grade 9 External Exam Papers and Bestseller Lists
- 5. Accessing Natural Science Grade 9 External Exam Papers Free and Paid eBooks
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- 6. Navigating Natural Science Grade 9 External Exam Papers eBook Formats
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  - Natural Science Grade 9 External Exam Papers Enhanced eBook Features
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  - o Adjustable Fonts and Text Sizes of Natural Science Grade 9 External Exam Papers
  - Highlighting and Note-Taking Natural Science Grade 9 External Exam Papers
  - Interactive Elements Natural Science Grade 9 External Exam Papers
- 8. Staying Engaged with Natural Science Grade 9 External Exam Papers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Natural Science Grade 9 External Exam Papers
- 9. Balancing eBooks and Physical Books Natural Science Grade 9 External Exam Papers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Natural Science Grade 9 External Exam Papers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Natural Science Grade 9 External Exam Papers
  - Setting Reading Goals Natural Science Grade 9 External Exam Papers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Natural Science Grade 9 External Exam Papers
  - Fact-Checking eBook Content of Natural Science Grade 9 External Exam Papers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements

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