Summative Test in Grade 10 Science Name: Score Name of Teacher Section: Instruction: Write the letter of the correct answer before the number. 1. Chemical reactions C. create and destroy atoms A. Occur only in living things B. Produce new substances D. only occur outside living organisms What are the coefficients that will balance the skeleton equation? AlCl3 + NaOH → Al(OH)3 + NaCil B. 3, 1, 3, 1 C. 1, 1, 1, 3 D. 1, 3, 3, 1 3 In every balanced chemical equation, each side of the equation has the same number of C moles D coefficients A. Atoms in each element B. molecules 4. Which molecules are most closely related to lipids? A. nucleotides B. fatty acids C. amino acida D. sugars 5. Which of the following molecules are classified as carbohydrates? A. mucleotides B. fatty acids C. amino acida D. sugars 6. Polysaccharides are \_ A. limids B. carbohydrates D. mucleic acids C. proteins 7. Which molecules are most closely related to proteins? A nucleotides B. fatty acids C. smano acids D. sugars 8. Which of the following is not an example of matter? B. heat C smodee D. water vapor 9. When the equation Fe + Cl2 -> FeCl3 is balanced, what is the coefficient for Cl2 ? A. 1 C: 3 D 4 10. What are the missing coefficients for the skeleton equation below? Al2 (SO4 )3 (aq) KOH(aq) → Al(OH)3 (aq) K2 SO4 (aq) B. 2,12,4,6 C 4.6.2.3 A. 1.3.2.3 11 The equation above is an example of what type of chemical reaction? A. combination B. decomposition C. single displacement D. double displacement. 12. The type of reaction that takes place when one element reacts with a compound to form a new compound and a different element is a reaction. B. decomposition A. combination C. single displacement D. double displacement. 13 Which of the following is a balanced equation representing the decomposition of lead(IV) oxide? C. Pb2O → 2Pb + O A P6O2 → P6 + 2O B. PbO2 → Pb + O2 D PbO → Pb + O2 14 Boyle's law states that A the volume of a gas varies inversely with pressure B. the volume of a gas varies directly with pressure C. the temperature of a gas varies inversely with pressure D. the temperature of a gas varies directly with pressure 15. Charles's law states that A. the pressure of a gas is inversely proportional to its temperature in kelvins B. the volume of a gas is directly proportional to its temperature in kelvins C. the pressure of a gas is directly proportional to its temperature in kelvins D. the volume of a gas is inversely proportional to its temperature in kelvins 16 As the temperature of a fixed volume of a gas increases, the pressure will A. Vary inversely C. not change B. decrease D. increase 17. A chemical reaction is a process in which: A. products change into reactants C. substances change state B. the law of conservation of mass applies D. all of these 18 During a chemical reaction; A. new elements are produced C. atoms are destroyed B. atoms are rearranged D. elements are destroyed 19. An equation is balanced by: A. changing subscripts C . erasing elements as necessary

D. adding elements as necessary

B. adding coefficients

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