# Hatchet by Gary Paulsen

## Reading comprehension questions

1. Who is the main character of "Hatchet"?
2. Where does the protagonist, Brian Robeson, travel to at the beginning of the story?
3. What is the reason for Brian's journey?
4. What vehicle does Brian use for his journey?
5. What happens to the pilot of the plane?
6. Where does Brian crash-land?
7. What is the main tool that helps Brian survive in the wilderness?
8. What does Brian call the survival tool he finds in the wreckage?
9. What type of food does Brian initially struggle to find in the wilderness?
10. How does Brian initially feel about being alone in the wilderness?
11. What kind of animal does Brian encounter early in the story?
12. How does Brian make fire for the first time?
13. What does Brian use to create sparks for the fire?
14. What does Brian use to catch fish?
15. How does Brian create a shelter?
16. What type of berries does Brian find and eat?
17. What is the biggest challenge Brian faces in terms of food?
18. What does Brian do to signal passing planes?
19. How does Brian deal with the mosquitoes in the wilderness?
20. What does Brian realize about his attitude towards self-pity?

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**Peipei Pang**

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web ksp for calcium hydroxide calcium hydroxide is a soft white caustic powder used in making mortar cements calcium salts paints and petrochemicals it is also used in saltwater aquaria to make up kalkwasser limewater solutions for reef tanks and is used as a ph regulating agent notice that calcium hydroxide is divalent and thus has twice

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web use the concentration of calcium hydroxide to calculate the concentration of calcium ions and hydroxide ions in the solution ca oh 2 à ca 2 2oh 5 x 10 2 m calcium ions 5 x 10 2 m 2 0 m hydroxide ions calculate calcium hydroxide ksp using the concentrations of calcium ions and hydroxide ions

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web madison bowe elana keller april 15 2014 introduction the objective of this lab was to measure the ksp of a slightly soluble basic salt and how solubility changes with temperature ksp is another form of equilibrium expression that deals with solubility the reaction that will take place is ca oh 2 ca2 2 oh

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web the chemical equation used was ca oh 2 s ca2 aq 2oh aq the solubility product was determined to be ksp ca2 oh from the titration which was made by dripping a concentrated solution of calcium hydroxide into hydrochloric acid the calcium hydroxide was filtered before starting the titration

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web lab report 8 determinin of calcium hydroxide the purpose of this was to titrat e a saturated ca oh 2 soultion with a st andard hcl solution and find the ksp or the oh concentration was f ound and then the concentration was found and u for this experiment a titration of calcium hydroxide with hydrochlo

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web therefore it was determined through error calculations and ease of usethat the most reliable valid method to use to get the closest ksp value tothe theoretical ksp for ca was the determination through ph method 2 introduction calcium hydroxide is a strong base soluble in water

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