



# Memorex Voice Recorder Manual

**Tobias Bleicker**



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mass 3 total mass of all atoms total of atoms

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web isotopes and weighted average atomic mass amu atomic mass unit mass of one proton mass of one neutron a sample of cesium cs has the following abundance cs 132 20 0 cs 133 75 3 cs 134 4 7 weighted average atomic mass  $132 \times 0.200 + 133 \times 0.752 + 134 \times 0.047 = 132.85$  amu

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web beanium isotope lab section 1 pre lab questions define average atomic mass write a mathematical equation that shows how you would determine the average atomic mass of an element section 2 data table don't forget calibration and units total all beans 100 beanium isotope blackium beanium isotope brownium beanium

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web beanium isotope b brownium beanium isotope c whiteium of atoms beans mass beans in cup mass empty cup mass of atoms beans 2 minus 3 average mass of 1 bean 4 divided by 1 of beans 1 a b c  $\times 100$  section 3 calculate the average atomic mass of beanium

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web in the beanium lab students obtained a sample of beanium a fictitious element in order to learn about isotopes and average atomic mass beanium has three isotopes white brown and speckled which have different masses and different percent abundances

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web 1 obtain a sample of beanium separate the pintonium navium and lentilium and then mass each group record the values in the data table 2 count the number of each isotope by counting the pintonium navium and lentilium 3 calculate the total number of isotopes

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web good luck safety 1 do not eat the beans 2 do not throw the beans 3 do not crush the beans don't mess with the beans pre lab questions 1 there are many beans in your container each bean represents an 2 there are three kinds of beans in your container each kind represents a different

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