



Classifying Chemical Reactions

Analyzing and Predicting Products

Introduction

The power of chemical reactions to transform our fives is visible all around us—in our houses, in our care, even in our bodies. Chemists try to make sense of the great variety of chemical reactions the same way that biologists organise their knowledge of life, by senting reactions into groups and classifying there. Classifying chemical reactions allows us to predict what chemical reactions will occur when different substances are missed.

Concepts

Chenical reactions

- Condination vs. decomposition
- Sinute ou double replacement.
- Condination rections

Background

A chemical reaction is defined as any process in which one or more substances are converted trainers with different properties. Chemical reactions change the identity of the reacting substances) and produce new substances. Observing the properties of the reactions and produces is therefore a key step to identifying chemical reactions. Some of the observations that may be associated with a chemical reaction include: (1) release of a gas, (2) formation of a precipitate; (3) color changes; (4) temperature changes; and (5) emission or absorption of light. As these observations suggest, thereford reactions can be dynamic and exciting events. The essence of any chemical reaction—reactions being transference into products—in summerical reaction.

The reasonate—more specifically, their formulas—are written on the left, side of the ecuation, the products on the right side of the ecuation. An arms represents the direction of the reastion and is read as "yields" or "produces." Other symbols are used to describe the physical state of the reaction and products and the reaction conditions (see Table 1).

Table 1. Symbols in Chemical Equations

| Syrabol | Translation |
|---------------|--|
| \rightarrow | Yields or produces (separates reactants from products) |
| * | Results with or forms alongwide frequencies two or noise resultants or producted |
| Δ | Reaction relation to be stell (well-tion over the arrow) |
| NA | No reaction takes place when reactions are of sec. |
| 140 | Pure substance freactant or productly is a solid- |
| Ob | Pure substance freactant or produced is a liquid |
| fact | Pure substance frequencies product is a gas |
| (ec) | As serves solution frequency or product is dissolved in water? |
| Cat | Catalyst—a substance needed to initiate a reaction Discougla is written over the arrows |

Chemical reactions may be classified by considering the number of reactants and products in the reaction, the physical or chemical nature of the reactants and products, and the reactangement of atoms in the conversion of the reactants into products (see Table 2 on page 2)

Experiment Overview

The purpose of this experiment is to observe a variety of chemical reactions and to identify putterns in the convention of reactants into products. The properties of the reactions will be analysed to classify the chemical reactions into different groups.

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PT Brinkman

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