

- Directions:**
 1. An element may be assigned a positive or negative charge depending on whether it is a metal or nonmetal.
 2. An atom with a positive charge is a cation.
 3. An atom with a negative charge is an anion.
 4. The number of protons in an atom is the atomic number.

Ion Practice Set

1. Write the symbol for the ion that has the same number of protons as the element but a different number of electrons.
 2. Write the symbol for the ion that has the same number of protons as the element but a different number of electrons.

Complete the following table, using the periodic table in the back of your book.

	ELEMENT NAME	ION SYMBOL	NUMBER OF PROTONS	NUMBER OF ELECTRONS	CHARGE OF ION (POSITIVE OR NEGATIVE)	TYPE OF ION
1	Sodium	Na^+	11	10	positive	cation
2	Chlorine	Cl^-	17	18	negative	anion
3	Strontium	Sr^{2+}	38	36	positive	cation
4	Fluorine	F^-	9	10	negative	anion
5	Calcium	Ca^{2+}	20	18	positive	cation
6	Oxygen	O^{2-}	8	10	negative	anion
7	Aluminum	Al^{3+}	13	10	positive	cation
8	Hydrogen	H^+	1	0	positive	cation
9	Lithium	Li^+	3	2	positive	cation
10	Magnesium	Mg^{2+}	12	10	positive	cation
11	Sulfur	S^{2-}	16	18	negative	anion
12	Selenium	Se^{2-}	34	36	negative	anion
13	Hydrogen	H^+	1	0	positive	cation
14	Boron	B^{3+}	5	2	positive	cation
15	Strontium	Sr^{2+}	38	36	positive	cation
16	Chlorine	Cl^-	17	18	negative	anion

Ions And Isotope Practice Set Answer Key

Patrick Vollmar



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