

IONIC AND METALLIC BONDING

Homework

Set 1: IONS

- For each element below, state (i) the number of valence electrons in the atom, (ii) the electron dot structure, and (iii) the chemical symbol(s) for the most stable ion.
 - Ba
 - I
 - K
- How many valence electrons does each of the following atoms have?
 - gallium
 - fluorine
 - selenium
- Write the electron configuration for each of the following atoms and ions.
 - Ca
 - chlorine atom
 - Na^+
 - phosphide ion
 - O^{2-}
- What is the relationship between the group number of the representative elements and the number of valence electrons?
- How many electrons will each element gain or lose in forming an ion? State whether the resulting ion is a cation or an anion.
 - strontium
 - aluminum
 - tellurium
 - rubidium
 - bromine
 - phosphorus
- Give the name and symbol of the ion formed when
 - a chlorine atom gains one electron.
 - a potassium atom loses one electron.
 - an oxygen atom gains two electrons.
 - a barium atom loses two electrons.
- How many electrons are lost or gained in forming each of the following ions?
 - Mg^{2+}
 - Br^-
 - Ag^+
 - Fe^{3+}
- Classify each of the following as a cation or an anion.
 - Na^+
 - Cu^{2+}
 - I^-
 - O^{2-}
 - Ca^{2+}
 - Cs^+

Set 2: IONIC BONDS AND IONIC COMPOUNDS

- Use electron dot structures to predict the formula of the ionic compounds formed when the following elements combine.
 - sodium and bromine
 - sodium and sulfur
 - calcium and iodine
 - aluminum and oxygen
 - barium and chlorine
- Which of these combinations of elements are most likely to react to form ionic compounds?
 - sodium and magnesium
 - barium and sulfur
 - potassium and iodine
 - oxygen and argon
- What is the meaning of coordination number?
- How is the coordination number determined?

Ionic And Metallic Bonding Test A Answers

SJ Ball



Ionic And Metallic Bonding Test A Answers:

Ionic And Metallic Bonding Test A Answers Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the ability of words has become more evident than ever. They have the ability to inspire, provoke, and ignite change. Such may be the essence of the book **Ionic And Metallic Bonding Test A Answers**, a literary masterpiece that delves deep to the significance of words and their affect our lives. Compiled by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall impact on readers.

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