

**DAV PUBLIC SCHOOL
KAILASH HILLS**

Max Marks 70

Time 3 Hrs

Session - 2014-15

FIRST TERMINAL EXAMINATION CLASS XII

Sub :- Chemistry

General Instructions

- | | |
|--------------------|--------------|
| Question No 1- 5 | 1 Mark Each |
| Question No 6- 10 | 2 Marks Each |
| Question No 11- 22 | 3 Marks Each |
| Question No 23 | 4 Marks Each |

USE LOG TABLE FOR CALCULATION

- Q No.1) What is the effect of the presence of Schottky defects on the density of a Crystal?
- Q No.2) A group of 14 elements is to be converted into n type semiconductor by doping it with a suitable impurity. To which group these impurities belong?
- Q No.3) 10 CC of a liquid A mixed with 10 cc liquid B. The volume of resulting solution was found to be 19.9 cc. What do you conclude?
- Q No.4) Can a reaction have zero activation energy?
- Q No.5) Explain why Cu^+ ion is not stable in aqueous solution?
- Q No.6) A Coordination compound has the formula, $\text{CoCl}_3 \cdot 4\text{NH}_3$. It does not liberate ammonia but precipitates one chloride ion as silver Chloride. Give the IUPAC name of the complex & write its structural formula.
- Q No.7) For the First order reaction, show that the time required for

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Karin Nielsen-Saines



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