

CHEM 121: Final Exam Study Guide

Chapter 1

- Know the Scientific method
- Know the definitions for hypothesis, scientific law, and scientific theory.

Chapter 2

- Length, mass, weight, volume
 - Know $1\text{ cm}^3=1\text{ mL}$ and $1\text{ dm}^3=1\text{ L}$
- Significant figures and Scientific notation
 - Rounding, in addition or subtraction, in multiplication or division
 - Keep as many sig figs until final answer
- Scientific notation
- Metric system
 - Know the metric prefixes: kilo (k), deci (d), centi (c), milli (m), and micro (μ)
 - Be able to perform metric-metric conversions using these prefixes.
- Use the metric-English conversions provided ($1\text{ in.}=2.54\text{ cm}$; $1\text{ lb}=453.6\text{ g}$; $1\text{ qt}=946\text{ mL}$)
- Volume by calculation
 - $V_{\text{rectangular solid}} = \text{length} \times \text{width} \times \text{thickness}$
- Volume by displacement
- Density: $d = \frac{m}{V}$
 - Calculate density, mass, or volume
 - Identify what items sink or float given densities of liquids and solids.
- Temperature
 - Know the formulas for converting "F-to-"C or "C-to-"F and K-to-"C or "C-to-K
- Percentage: ratio of parts per 100 parts
 - Given amount of part and whole, calculate %
 - Use a given % to solve for part or whole

Chapter 3

- Know that matter is studied at the macroscopic, microscopic, particulate (molecular) levels
- Physical states of matter
 - Determine physical state of substances (solids, liquids, gases) given descriptions of volume, shape, particles moving, etc.
- Classify properties and changes as physical or chemical
 - Know terms for changes of state:
 - Melting, freezing, vaporizing, condensation, sublimation, deposition
- Classification of matter
 - Given examples, determine which are elements, compounds, or mixtures
 - Given molecular-level images, determine which are elements, compounds, or mixtures and solids, liquids, or gases
 - Distinguish between homogeneous and heterogeneous mixtures
- Chemical reaction:
 - reactants: starting materials
 - products: substances produced in reaction

Final Exam Study Guide Chemistry

Yicheng Fang



Final Exam Study Guide Chemistry:

This book delves into Final Exam Study Guide Chemistry. Final Exam Study Guide Chemistry is an essential topic that must be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Final Exam Study Guide Chemistry, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to Final Exam Study Guide Chemistry
 - Chapter 2: Essential Elements of Final Exam Study Guide Chemistry
 - Chapter 3: Final Exam Study Guide Chemistry in Everyday Life
 - Chapter 4: Final Exam Study Guide Chemistry in Specific Contexts
 - Chapter 5: Conclusion
2. In chapter 1, the author will provide an overview of Final Exam Study Guide Chemistry. This chapter will explore what Final Exam Study Guide Chemistry is, why Final Exam Study Guide Chemistry is vital, and how to effectively learn about Final Exam Study Guide Chemistry.
3. In chapter 2, this book will delve into the foundational concepts of Final Exam Study Guide Chemistry. This chapter will elucidate the essential principles that must be understood to grasp Final Exam Study Guide Chemistry in its entirety.
4. In chapter 3, the author will examine the practical applications of Final Exam Study Guide Chemistry in daily life. The third chapter will showcase real-world examples of how Final Exam Study Guide Chemistry can be effectively utilized in everyday scenarios.
5. In chapter 4, the author will scrutinize the relevance of Final Exam Study Guide Chemistry in specific contexts. This chapter will explore how Final Exam Study Guide Chemistry is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Final Exam Study Guide Chemistry. The final chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Final Exam Study Guide Chemistry.

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