

- Directions:**
 1. An element may be assigned a positive or negative charge depending on whether it is a metal or nonmetal.
 2. An atom with a positive charge is a cation.
 3. An atom with a negative charge is an anion.
 4. The number of protons in an atom is the atomic number.

Ion Practice Set

1. Write the symbol for the ion that has the following number of protons and electrons.
 2. Write the symbol for the ion that has the following number of protons and electrons.
 3. Write the symbol for the ion that has the following number of protons and electrons.

Complete the following table, using the periodic table in the back of your book.

	Element Name	Ion Symbol	Number of Protons	Number of Electrons	Charge on Ion (lost or gained)	Type of Ion
1	Sodium	Na^+	11	10	lost one	cation
2	Chlorine	Cl^-	17	18	gained one	anion
3	Calcium	Ca^{2+}	20	18	lost two	cation
4	Fluorine	F^-	9	10	gained one	anion
5	Magnesium	Mg^{2+}	12	10	lost two	cation
6	Oxygen	O^{2-}	8	10	gained two	anion
7	Hydrogen	H^+	1	0	lost one	cation
8	Lithium	Li^+	3	2	lost one	cation
9	Boron	B^{3+}	5	2	lost three	cation
10	Carbon	C^{4-}	6	10	gained four	anion
11	Nitrogen	N^{3-}	7	10	gained three	anion
12	Fluorine	F^-	9	10	gained one	anion
13	Sulfur	S^{2-}	16	18	gained two	anion
14	Chlorine	Cl^-	17	18	gained one	anion
15	Argon	Ar^{0}	18	18	lost none	neutral
16	Helium	He^{0}	2	2	lost none	neutral
17	Sodium	Na^+	11	10	lost one	cation
18	Chlorine	Cl^-	17	18	gained one	anion

Ion Practice Set Answers

Christian Drosten

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

Ion Practice Set Answers:

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