

- Directions:**
 1. An element may be assigned a positive or negative charge depending on whether it is a metal or nonmetal.
 2. An atom with a positive charge is a cation.
 3. An atom with a negative charge is an anion.
 4. The number of protons in an atom is the atomic number.

Ion Practice Set

1. Write the symbol for the ion that has the following number of protons and electrons.
 2. Write the symbol for the ion that has the following number of protons and electrons.
 3. Write the symbol for the ion that has the following number of protons and electrons.

Complete the following table, using the periodic table in the back of your book.

	Element Name	Ion Symbol	Number of Protons	Number of Electrons	Charge on Ion	Type of Ion
1	Sodium	Na^+	11	10	+1	Cation
2	Chlorine	Cl^-	17	18	-1	Anion
3	Calcium	Ca^{2+}	20	18	+2	Cation
4	Oxygen	O^{2-}	8	10	-2	Anion
5	Aluminum	Al^{3+}	13	10	+3	Cation
6	Sulfur	S^{2-}	16	18	-2	Anion
7	Hydrogen	H^+	1	0	+1	Cation
8	Lithium	Li^+	3	2	+1	Cation
9	Fluorine	F^-	9	10	-1	Anion
10	Magnesium	Mg^{2+}	12	10	+2	Cation
11	Strontium	Sr^{2+}	38	36	+2	Cation
12	Barium	Ba^{2+}	56	54	+2	Cation
13	Iron	Fe^{2+}	26	24	+2	Cation
14	Copper	Cu^+	29	28	+1	Cation
15	Gold	Au^+	79	78	+1	Cation
16	Mercury	Hg^{2+}	80	78	+2	Cation
17	Lead	Pb^{2+}	82	80	+2	Cation
18	Plutonium	Pu^{4+}	94	90	+4	Cation
19	Uranium	U^{4+}	92	88	+4	Cation
20	Neptunium	Np^{3+}	93	90	+3	Cation
21	Polonium	Po^{2-}	84	86	-2	Anion
22	Astatine	At^-	85	86	-1	Anion
23	Francium	Fr^+	87	86	+1	Cation
24	Radium	Ra^{2+}	88	86	+2	Cation
25	Actinium	Ac^+	89	88	+1	Cation

Ion And Isotope Practice Set Answer Key

Siddappa N.Byrareddy



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