

# **Naming Compounds Practice Problems** Naming Compounds: Select from the following list: Acid, Base, Binary Compound, Ionic Compound, Molecular Compound, Organic Compound, or Other.

Name the following compounds:

- 1)  $\text{HNO}_3$ : **nitric acid**
- 2)  $\text{NaOH}$ : **sodium hydroxide**
- 3)  $\text{H}_2\text{SO}_4$ : **sulfuric acid**
- 4)  $\text{NaCl}$ : **sodium chloride**
- 5)  $\text{H}_2\text{O}$ : **water**
- 6)  $\text{H}_2\text{SO}_3$ : **sulfurous acid**
- 7)  $\text{HNO}_2$ : **nitrous acid**
- 8)  $\text{H}_2\text{S}$ : **hydrogen sulfide**
- 9)  $\text{Na}_2\text{SO}_4$ : **sodium sulfate**
- 10)  $\text{NaOH}$ : **sodium hydroxide**
- 11)  $\text{H}_2\text{SO}_4$ : **sulfuric acid**

Write the formula for the following compounds:

- 1) sodium chloride:  **$\text{NaCl}$**
- 2) ammonium sulfate:  **$(\text{NH}_4)_2\text{SO}_4$**
- 3) nitric acid:  **$\text{HNO}_3$**
- 4) acetic acid:  **$\text{CH}_3\text{COOH}$**
- 5) phosphorus pentachloride:  **$\text{PCl}_5$**
- 6) silver chloride:  **$\text{AgCl}$**
- 7) sodium nitrate:  **$\text{NaNO}_3$**
- 8) copper(II) sulfate:  **$\text{CuSO}_4$**
- 9) calcium chloride:  **$\text{CaCl}_2$**
- 10) sodium hydroxide:  **$\text{NaOH}$**
- 11) phosphorus pentachloride:  **$\text{PCl}_5$**

# Hs Science Unit 10 Lesson 1 Answeres

**Lingsheng Yao**

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