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**ML Yell**



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web this is like a one dimensional system whose mechanical energy  $e$  is a constant and whose potential energy with respect to zero energy at zero displacement from the spring  $s$  unstretched length  $x = 0$  is  $u = \frac{1}{2} kx^2$  figure 8 5 2 8 5 2 a a glider between springs on an air track is an example of a horizontal mass spring system

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web 1 if the  $pe$  of the reactants is 430kJ and the  $pe$  of the products is 190kJ what is the  $\Delta h$  2 if the  $pe$  of the reactants is 430kJ and the activation energy is 220 how much total energy will be needed to be added for the reaction to proceed transcribed image text potential energy diagram answer the questions using the graph

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web answer see below explanation the rate of a chemical reaction  $a + b \rightarrow c + d$  is given in first approximation by  $v = k_v \frac{a}{b}$  where  $k_v = k e^{-E_a/RT}$  where  $k$  depends by many factors as solvent geometry type of reaction  $E_a$  is the activation energy  $R$  is the universal constant of gases  $T$  is the absolute temperature

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web may 1 2013 a potential energy diagram shows the change in potential energy of a system as reactants are converted into products figure below shows basic potential energy diagrams for an endothermic a and an exothermic b reaction recall that the enthalpy change  $\Delta h$  is positive for an endothermic reaction and negative for an

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