

## Chapter Test B

### Chapter: Chemical Equations and Reactions

Write the full name (including unit) for each of the units in places that have one. Place each statement in the correct column with question.

1. The coefficient of a chemical equation indicates the number of molecules of each substance that are involved in the reaction.
  - a. gram
  - b. grams/mole
  - c. molecules/molecule
  - d. moles/mole
2. The coefficient of a chemical equation is a percentage of the whole.
  - a. molecules
  - b. molecules
  - c. molecules/mole
  - d. number of products
3. An unbalanced equation does not tell you anything about the substances in it.
  - a. matter
  - b. chemical equation
  - c. matter
  - d. molecules
4. The coefficient of a chemical equation indicates the number of molecules of each substance that are involved in the reaction.
  - a. molecules/mole
  - b. molecules/mole
  - c. molecules/mole
  - d. molecules/mole
5. The coefficient of a chemical equation indicates the number of molecules of each substance that are involved in the reaction.
  - a. molecules/mole
  - b. molecules/mole
  - c. molecules/mole
  - d. molecules/mole
6. The coefficient of a chemical equation indicates the number of molecules of each substance that are involved in the reaction.
  - a. molecules/mole
  - b. molecules/mole
  - c. molecules/mole
  - d. molecules/mole
7. The coefficient of a chemical equation indicates the number of molecules of each substance that are involved in the reaction.
  - a. molecules/mole
  - b. molecules/mole
  - c. molecules/mole
  - d. molecules/mole
8. The coefficient of a chemical equation indicates the number of molecules of each substance that are involved in the reaction.
  - a. molecules/mole
  - b. molecules/mole
  - c. molecules/mole
  - d. molecules/mole

# Modern Chemistry Ch 11 Test B Answers

**Joan I. Senyk, Darrell D. Ebbing**



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