

Chemical Bonding

Section 1: Molecular Geometry

For this section, please select the letter of the best answer to each question. Questions that have multiple correct answers will be marked as such.

1. Which of the following is true for a molecule with a central atom bonded to four other atoms?
 - a. The molecule is always tetrahedral.
 - b. It is always tetrahedral.
 - c. The molecule is always tetrahedral if the central atom has no lone pairs.
 - d. The molecule is always tetrahedral.
2. Which of the following is true for a molecule with a central atom bonded to three other atoms?
 - a. The molecule is always trigonal planar.
 - b. The molecule is always trigonal planar.
 - c. The molecule is always trigonal planar if the central atom has no lone pairs.
 - d. The molecule is always trigonal planar.
3. Which of the following is true for a molecule with a central atom bonded to two other atoms?
 - a. The molecule is always linear.
 - b. The molecule is always linear.
 - c. The molecule is always linear if the central atom has no lone pairs.
 - d. The molecule is always linear.
4. Which of the following is true for a molecule with a central atom bonded to one other atom?
 - a. The molecule is always linear.
 - b. The molecule is always linear.
 - c. The molecule is always linear if the central atom has no lone pairs.
 - d. The molecule is always linear.
5. Which of the following is true for a molecule with a central atom bonded to no other atoms?
 - a. The molecule is always linear.
 - b. The molecule is always linear.
 - c. The molecule is always linear if the central atom has no lone pairs.
 - d. The molecule is always linear.

Modern Chemistry Chemical Bonding Section Quiz

Molecular Geometry Answers

J Ma

A decorative graphic consisting of a light blue horizontal bar with a rounded right end, and a red circular shape partially visible behind it.

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