

- (i) It depends on the concentration of a non-electrolyte solute in a solution and does not depend on the nature of the solute molecules.
- (ii) It depends on a number of particles of electrolyte solute in a solution and does not depend on the nature of the solute particles.
- (iii) It depends on the concentration of a non-electrolyte solute in solution as well as on the nature of the solute molecules.
- (iv) It depends on the concentration of an electrolyte or nonelectrolyte solute in solution as well as on the nature of solute molecules.

**Solution:**

Option (i) and (ii) are the answers.

**30. Van't Hoff factor  $i$  is given by the expression \_\_\_\_\_.**

- (i)  $i = \text{Normal molar mass} / \text{Abnormal molar mass}$
- (ii)  $i = \text{Abnormal molar mass} / \text{Normal molar mass}$
- (iii)  $i = \text{Observed colligative property} / \text{calculated colligative property}$
- (iv)  $i = \text{Calculated colligative property} / \text{Observed colligative property}$

**Solution:**

Option (i) and (iii) are the answers.

**31. Isotonic solutions must have the same \_\_\_\_\_.**

- (i) solute
- (ii) density
- (iii) elevation in boiling point
- (iv) depression in freezing point

**Solution:**

Option (iii) and (iv) are the answers.

**32. Which of the following binary mixtures will have the same composition in the liquid and vapour phase?**

- (i) Benzene - Toluene
- (ii) Water-Nitric acid
- (iii) Water-Ethanol
- (iv) *n*-Hexane - *n*-Heptane

**Solution:**

Option (ii) and (iii) are the answers.

**33. In isotonic solutions \_\_\_\_\_.**

- (i) solute and solvent both are same.
- (ii) osmotic pressure is the same.
- (iii) solute and solvent may or may not be same.
- (iv) a solute is always the same solvent may be different.

**Solution:**

Option (ii) and (iii) are the answers.

**34. For a binary ideal liquid solution, the variation in total vapour pressure versus the composition of the solution is given by which of the curves?**

# Modern Chemistry Chapter 12 Section 2 Review Answers

**M Tight**



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**Food Analysis** S. Suzanne Nielsen,2017-06-06 This fifth edition provides information on techniques needed to analyze foods for chemical and physical properties The book is ideal for undergraduate courses in food analysis and is also an invaluable reference to professionals in the food industry General information chapters on regulations labeling sampling and data handling provide background information for chapters on specific methods to determine chemical composition and characteristics physical properties and objectionable matter and constituents Methods of analysis covered include information on the basic principles advantages limitations and applications Sections on spectroscopy and chromatography along with chapters on techniques such as immunoassays thermal analysis and microscopy from the perspective of their use in food analysis have been expanded Instructors who adopt the textbook can contact the editor for access to a website with related teaching materials **India Rubber World and Electrical Trades Review** ,1927 Modern Aspects of

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*Pharmaceutical Journal*, 1961

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