

CHEMISTRY

CHEAT SHEET

CHEMICAL KINETICS

CHEMICAL KINETICS CHEAT SHEET

DEFINITIONS: The study of the rates of chemical reactions is called chemical kinetics. It is concerned with the rates at which chemical reactions take place and the factors that affect the rates of chemical reactions.

Rate of Reaction: The rate of reaction is the change in concentration of a reactant or product per unit time.

Factors Affecting the Rate of Reaction:

1. Concentration of reactants
2. Temperature
3. Surface area of solid reactants
4. Presence of a catalyst

Rate Law: A mathematical expression that relates the rate of a reaction to the concentrations of the reactants.

Order of Reaction: The sum of the powers of the concentration terms in the rate law is called the order of reaction.

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Order	Rate Law	Half-life	Graph
Zero order	$\text{Rate} = k[A]^0 = k$	$t_{1/2} = \frac{[A]_0}{k}$	
First order	$\text{Rate} = k[A]^1$	$t_{1/2} = \frac{0.693}{k}$	
Second order	$\text{Rate} = k[A]^2$	$t_{1/2} = \frac{1}{k[A]_0}$	

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Susan Rozzi, Michelle Futrell

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