

Assessment

Quiz

Section: Describing Chemical Reactions

In the space provided, write the letter of the term or phrase that best answers the question.

- _____ 1. The formation of a gas is evidence of
a. a chemical change.
b. a physical change.
c. no change in energy.
d. Both (a) and (c).
- _____ 2. Which of the following may be evidence of a chemical change?
a. release of energy as heat
b. production of sound
c. reduction in temperature
d. All of the above
- _____ 3. In the word equation,
iron + copper(II) sulfate $\rightarrow$ iron(II) sulfate + copper,
a product is
a. iron(II) sulfate,
b. copper,
c. iron,
d. Both (a) and (b)
- _____ 4. In the reaction described by the following equation
water + energy $\rightarrow$ hydrogen gas + oxygen gas,
energy
a. can be considered a reactant.
b. is absorbed.
c. is released.
d. Both (a) and (b)
- _____ 5. In the unbalanced formula equation
carbon + hydrogen $\rightarrow$ methane, the formula
 CH_4 represents
a. one atom of methane.
b. five atoms of methane.
c. one molecule of methane.
d. one gram of methane.

Holt Chemistry Chemical Equations And Reactions Answers

Tobias Bleicker



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