

Chapter Test B, continued _____

17. A substance combines with oxygen, releasing a large amount of energy as heat and light, as in: _____
18. The decomposition of a substance by an electric current is called _____.
19. A(n) _____ oxidizes the elements by the mass with which they combine, forming chemical reactions.
20. A(n) _____, upon exposure and the release of electrons, the masses and relative amounts of free ions and products in a chemical reaction.
21. The law of _____ means that not all that is chemical equation to be balanced.
22. A(n) _____ product is a substance indicated by an arrow pointing upward.
23. Two equalled arrows mean that the products are _____.

PART B Write the answers to the following questions in the space provided.

24. How are a word equation and a formula equation alike? How are they alike (How is chemical equation)?

25. What are three observations that indicate a chemical change may be taking place?

PART C Write a word equation for each chemical reaction in the space provided.



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J Rink



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