

Name: _____ DawgTag: _____

1. In the molecule methanol (COH_4), which atom will be in the center of the skeleton of the Lewis structure?

Carbon

2. What is the name given to multiple Lewis structures that describe the same molecule?

Resonance Structure

3. If two or more Lewis structures are possible to represent a molecule, what criterion is used to determine which structure or structures is preferred?

- smaller formal charge preferable to large.

- Like formal charge on adj atoms undesired (unless 0)

- more negative formal charge goes to more electronegative atom.

4. What is the shape of a molecule with three atoms bound and one lone pair of electrons on the central atom?

Trigonal Pyramidal

5. What is the shape of a molecule with the 2 lone pairs and 3 atoms bound to the central atom?

T-shaped

6. What is the shape of a CO_2 molecule?

Linear



7. What is the shape of a NH_3 molecule?

Trigonal Pyramid



8. If a central atom is making 5 bonds with 5 other atoms, what is the orbital hybridization of that central atom?

 sp^3d

9. If a central atom is making 3 bonds with 3 other atoms, what is the orbital hybridization of that central atom?

 sp^2

10. In the molecule methane, CH_4 , what is the H-C-H bond angle?

 109.5° 

Molecular Geometry And Bonding Pre Lab Answers

Clemens Wendtner



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