

CHEMISTRY MOLE COOKIES

Practice
mole
calculations
with a
cookie
recipe!



Chemistry Mole Cookies Worksheet

Instructions to complete the recipe for chocolate chip cookies:

1. Fill in the table below by first determining each ingredient's molecular mass.
2. Convert the ingredient amounts, which are listed in grams, then grams to the proper cooking units. (Use the conversions below.)
3. Complete the recipe card on the back.

Notes: Many of the ingredients are molecules, so have complete structural formulas. They have been simplified here.

INGREDIENT	MOLECULAR MASS <small>molecular mass</small>	MOLES <small>moles</small>	GRAMS <small>grams</small>	COOKING UNIT <small>amount?</small>
flour <chem>C6H12O6</chem>		1.50 moles		
baking soda <chem>NaHCO3</chem>		0.0050 moles		
salt <chem>NaCl</chem>		0.004 moles		
butter <chem>C4H8O2</chem>		1.000 moles		
white sugar <chem>C12H22O11</chem>		0.44 moles		
brown sugar <chem>C12H22O11</chem>		0.40 moles		
egg <chem>C6H12O6</chem>		0.024 moles		
vanilla <chem>C8H8O3</chem>		0.002 moles		
chocolate chips <chem>C42H58O6</chem>		1.40 moles		

Use the following conversions to get from grams to standard cooking units of measure.

- 1 cup flour = 125.00 grams
- 1 cup brown sugar = 225.00 grams
- 1 large egg = 50 grams
- 3 teaspoons vanilla = 4.75 grams
- 1 cup chocolate chips = 175.00 grams

- 1 cup white sugar = 200.00 grams
- 1 teaspoon baking soda = 0.44 grams
- 1 teaspoon salt = 0.50 grams
- 1 cup butter = 225.00 grams

continued on back

Mole Cookies Answers

Carolyn A. Fischer



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Lighting for Digital Video and Television John

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