

# MOLARITY BY DILUTION

79 + 52 = 131 Name \_\_\_\_\_

Acids are usually acquired from chemical supply houses in concentrated form. These acids are diluted to the desired concentration by adding water. Since moles of acid before dilution = moles of acid after dilution, and moles of acid =  $M \times V$  then,  
 $M_1 \times V_1 = M_2 \times V_2$  Solve the following problems.

1. How much concentrated 18 M sulfuric acid is needed to prepare 250 mL of a 6.0 M solution?

$$V_1 = \frac{M_2 V_2}{M_1} = \frac{(6.0 M)(250 mL)}{18 M}$$

83 mL

2. How much concentrated 12 M hydrochloric acid is needed to prepare 100 mL of a 2.0 M solution?

$$V_1 = \frac{M_2 V_2}{M_1} = \frac{(2.0 M)(100 mL)}{12 M}$$

17 mL

3. To what volume should 25 mL of 15 M nitric acid be diluted to prepare a 3.0 M solution?

$$V_2 = \frac{M_1 V_1}{M_2} = \frac{(15 M)(25 mL)}{3.0 M}$$

125 mL

4. To how much water should 50 mL of 12 M hydrochloric acid be added to produce a 4.0 M solution?

$$V_2 = \frac{M_1 V_1}{M_2} = \frac{(12 M)(50 mL)}{4.0 M} = 150 mL - 50 mL$$

100 mL H<sub>2</sub>O

5. To how much water should 100 mL of 18 M sulfuric acid be added to prepare a 1.5 M solution?

$$V_2 = \frac{M_1 V_1}{M_2} = \frac{(18 M)(100 mL)}{1.5 M} = 1200 mL - 100 mL = 1100 mL$$

1.1 L

# Molarity By Dilution Answer Key

**Carson-Dellosa Publishing**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the Carson-Dellosa Publishing text.

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**Techniques** Philip L.R. Bonner, Alan J. Hargreaves, 2011-08-24 This unique practical pocket sized guide and reference provides every first year bioscience student with all they need to know to prepare reagents correctly and perform fundamental laboratory techniques It also helps them to analyse their data and present their findings in addition to directing the reader via a comprehensive list of references to relevant further reading All of the core bioscience laboratory techniques are covered including basic calculations and the preparation of solutions aseptic techniques microscopy techniques cell fractionation spectrophotometry chromatography of small and large molecules electrophoresis of proteins and nucleic acids and data analysis In addition the book includes clear relevant diagrams and worked examples of calculations In short this is a must have for all first year bioscience students struggling to get to grips with this vitally important element of their course

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