

Experiment 3: Chemical Kinetics

Report Sheet/Post Lab Questions

Measure	Solution 1	Solution 2	Solution 3	† Temperature
Volume H ₂ O ₂ (mL)	5.0	10	5	5.0
Volume H ₂ O (mL)	15.0	10.0	5.0	15.0
Volume KI (mL)	10	10	20	10
Slope	0.1484	0.2677	0.3129	0.3018

1. Show your calculation for the slope.

$$m = 1$$

$$n = 2$$

2. Write the Rate Law for this reaction

$$\text{rate} = k[A][B]^2$$

3. What is the order of this reaction?

$$m+n = 3$$

4. Did the rate for the higher temperature increase or decrease?

higher

5. Look up the solubility of oxygen in water.

faster

6. From the solubility value, do you expect that its solubility can affect the experimental results? Explain why.

7. What are some possible errors in this experiment?

the time we get is no precise

8. Was iodide left in the reaction solution after it was allowed to go to completion? What can you conclude about iodide?

9. From the data collected could the activation energy for the reaction be calculated?

Kinetics Of A Reaction Lab Answers

M Mosston



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Scientific and Technical Aerospace Reports ,1986 Lists citations with abstracts for aerospace related reports obtained from world wide sources and announces documents that have recently been entered into the NASA Scientific and Technical Information Database **Reaction Rates for High-temperature Air with Carbon and Sodium Impurities** Mina L.

Carnicom,1968 The values used by a number of investigators for the rate constants of high temperature greater than or equal to 1000 K homogeneous gaseous reactions involving species of the elements nitrogen oxygen carbon and sodium have been compiled and are presented in tabular form Included are reactions involving neutral species charged species free electrons some species in excited electronic or vibrational states and radiative processes **American Laboratory** ,2003

Energy Research Abstracts ,1989 Semiannual with semiannual and annual indexes References to all scientific and technical literature coming from DOE its laboratories energy centers and contractors Includes all works deriving from DOE other related government sponsored information and foreign nonnuclear information Arranged under 39 categories e g Biomedical sciences basic studies Biomedical sciences applied studies Health and safety and Fusion energy Entry gives bibliographical information and abstract Corporate author subject report number indexes **Publications of the National**

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Current Catalog National Library of Medicine (U.S.),1968 Includes subject section name section and 1968 1970
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concerned with the demonstration of the applicability of techniques to a broad spectrum of materials and to establish the
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International Conference was unique in that the first standards were disclosed for differential thermal analysis This was the
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majority Realization of full maturity can be expected in the near future Inclusion of plenary lectures in these volumes
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expressed the hope that thermal methods of analysis would find wider application in science and technology The citation
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U S A President ICTA 1968 1971 XIII PREFACE For the past two decades thermo analytical methods have reached a stage of
considerable importance which is particularly due to the developments in the area of instrumentation *The Journal of Solid
Waste Technology and Management* ,1996

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