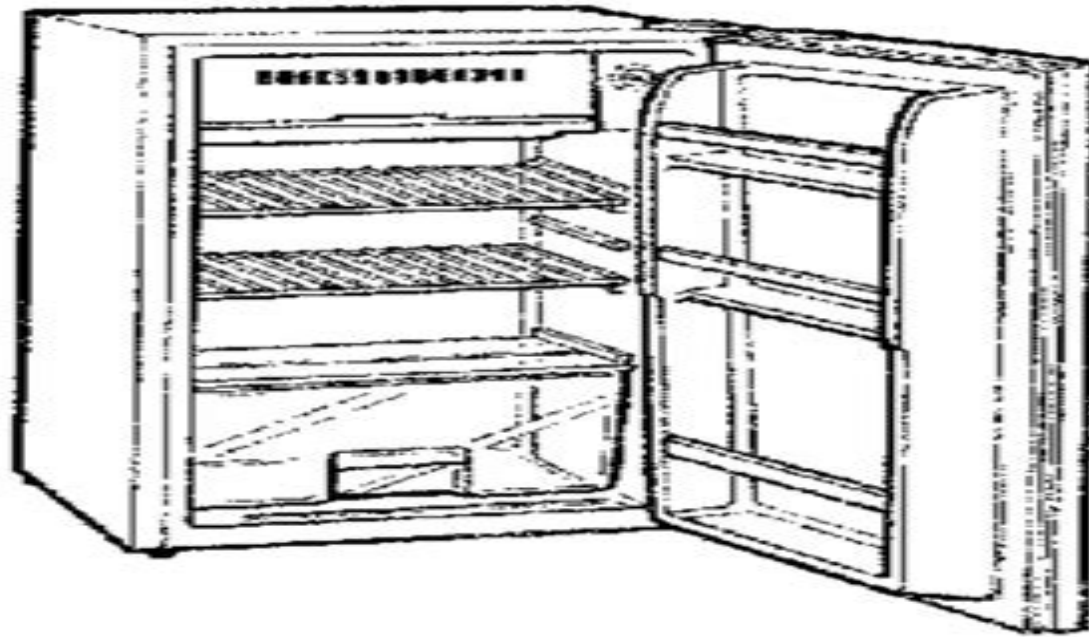




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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c

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web pressure increases even though there were errors in the experiment the goal of the experiment was met the goal of
declaring what type of relationship it is was clear using the data pressure kpa temperature c temperature k constant k p t 95
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condition at which a phase change occurs from liquid to vapour or vapour to liquid

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the thermometer and pressure gauge indicate the temperature and the pressure qualitatively the level in the flask indicates
the volume and the number of particles in each flask indicates relative amounts