

Chapter Test B

Chapter: Measurements and Calculations

PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. Which of the following is *not* a stage of the scientific method?
- using a hypothesis to sell a new product
 - observing, collecting, and communicating data
 - formulating theories
 - formulating hypotheses
- _____ 2. By making several measurements of a sample with the same balance, a chemist obtained values of 5.224 g, 5.235 g, and 5.250 g for the mass of a sample. Without knowing the actual mass of the sample, we can tell that these measurements have
- high precision.
 - high accuracy.
 - low precision.
 - low accuracy.
- _____ 3. Which equation represents an inverse proportion between the variables?
- $V = \frac{T}{k}$
 - $F = -kx^2$
 - $PV = k$
 - $y = kx - 8$
- _____ 4. The average distance between the sun and Earth is about 150 million km. Light travels at a speed of 3.0×10^8 m/s. Dividing the distance by the speed and then multiplying by the number of meters in a kilometer will result in units of
- seconds.
 - meters per second.
 - kilometers squared.
 - kilometers per second.
- _____ 5. In which of the following measurements are all the zeros considered to be nonsignificant figures?
- 506 mL
 - 60.0 mL
 - 0.0037 mL
 400. mL

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A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the text 'Holt, Rinehart and Winston Staff'.

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Holt Chemistry Ralph Thomas Myers, 2004 **Holt Chemistry File**, 1998 This reference is a must for students who need extra help reteaching or extra practice The guide moves students through the same concepts as the text but at a slower pace More descriptive detail along with visual algorithms provides a more structured approach Each chapter closes with a large bank of practice problems Book jacket Holt Biology Holt Rinehart & Winston, 2003-08 *Modern Chemistry* Holt Rinehart & Winston, 1998-01-01 2000 2005 State Textbook Adoption Rowan Salisbury Modern Chemistry Holt, Rinehart and Winston Staff, 2006-01 **Holt Physical** Holt, Rinehart and Winston Staff, 1994 *Children's Books in Print* R R Bowker Publishing, Bowker, 1999-12 **Advanced Dairy Chemistry: Volume 1: Proteins, Parts A&B** Patrick F. Fox, Paul L. H. McSweeney, 2013-09-24 Advanced Dairy Chemistry I Proteins is the first volume of the third edition of the series on advanced topics in Dairy Chemistry which started in 1982 with the publication of *Developments in Dairy Chemistry* This series of volume is intended to be a coordinated and authoritative treatise on Dairy Chemistry In the decade since the second edition of this volume was published 1992 there have been considerable advances in the study of milk proteins which are reflected in changes to this book All topics included in the second edition are retained in the current edition which has been updated and considerably expanded from 18 to 29 chapters Owing to its size the book is divided into two parts Part A Chapters 1 11 describes the more basic aspects of milk proteins while Part B Chapters 12 29 reviews the more applied aspects Chapter 1 a new chapter presents an overview of the milk protein system especially from an historical viewpoint Chapters 2 5 7 9 15 and 16 are revisions of chapters in the second edition and cover analytical aspects chemical and physiochemical properties biosynthesis and genetic polymorphism of the principal milk proteins Non bovine caseins are reviewed in Chapter 6 Trace Environmental Quantitative Analysis Paul R. Loconto, 2020-12-27 A thorough and timely update this new edition presents principles techniques and applications in this sub discipline of analytical chemistry for quantifying traces of potentially toxic organic and inorganic chemical substances found in air soil fish and water as well as serum plasma urine and other body fluids The author addresses regulatory aspects calibration verification and the statistical treatment of analytical data including instrument detection limits quality assurance quality control sampling and sample preparation and techniques that are used to quantify trace concentrations of organic and inorganic chemical substances Key Features Fundamental principles are introduced for the more significant experimental approaches to sample preparation Principles of instrumental analysis determinative techniques for trace organics and trace inorganics analysis An introduction to the statistical treatment of trace analytical data How to calculate instrument detection limits based on weighted least squares confidence band calibration statistics Includes an updated series of student tested experiments Holt General Science William L. Ramsey, 1988

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