

Name _____ Class _____ Date _____

Skills Worksheet

Concept Review

Section: How are Elements Organized?

Answer the following questions in the space provided.

1. Why do Li, Na, K, Rb, Cs, and Fr all react with Cl in a 1:1 ratio forming substances with similar properties?

2. Explain the method that John Newlands used to organize the elements.

3. What method did Dmitri Mendeleev use to arrange his periodic table?

4. Why did Mendeleev have gaps in his table? How did he use these gaps?

5. What was Henry Moseley's contribution to the periodic table?

Concept Review continued

6. Why was Moseley able to resolve the discrepancy Mendeleev could not?

7. Explain the importance of valence electrons.

8. Why do elements with similar properties appear in the same column in the periodic table?

9. How is the electron configuration similar for each element in a column?

10. How is the electron configuration similar for each element in a row?

Holt Chemistry Concept Review Trends Periodic Table Answers

Ying Liu



Holt Chemistry Concept Review Trends Periodic Table Answers:

Whispering the Techniques of Language: An Mental Journey through **Holt Chemistry Concept Review Trends Periodic Table Answers**

In a digitally-driven earth where screens reign great and immediate communication drowns out the subtleties of language, the profound techniques and mental subtleties hidden within words often go unheard. Yet, located within the pages of **Holt Chemistry Concept Review Trends Periodic Table Answers** a charming literary treasure pulsing with organic feelings, lies an extraordinary journey waiting to be undertaken. Published by a talented wordsmith, that charming opus attracts visitors on an introspective journey, delicately unraveling the veiled truths and profound impact resonating within the fabric of each word. Within the emotional depths of the moving evaluation, we will embark upon a honest exploration of the book is primary themes, dissect their fascinating writing style, and yield to the effective resonance it evokes strong within the recesses of readers hearts.

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