

## Ch. 11 Gas Laws Chapter Review Answers

1. Particles of matter are always in motion. Gas particles are small and have a huge amount of space between them compared to their size. Because they are so far apart, they have no attraction between each other, and they have highly elastic collisions. The temperature of the gas is the average kinetic energy of the gas particles.
2. An ideal gas is a theoretical gas that has perfect particle qualities that fit the molecular-kinetic theory of matter.
3. Pressure:
  - a. The fast moving gas particles strike the sides of the container they are in.
  - b. The same force applied to a smaller area will calculate to a greater pressure.
4. Mercury and Pressure:
  - a. The pressure of the mercury in the tube is equal to the atmospheric pressure at sea-level, which is 101.325 kPa, or 1 atm, or 760 mmHg.
  - b. 10.3 m, or approximately 34 feet.
  - c. Mercury's density is 13.5 times greater than the density of water, therefore, it exerts 13.5 times more pressure than a column of water the same size.
5. Units of Pressure:
  - a. mmHg, in.Hg, torr, bar, psi, atm., Pa
  - b. 1 atm = 760 mmHg
  - c. A Pascal is the force of 1 N exerted over an area of 1 m<sup>2</sup>. 1 Pa = 1N/1 m<sup>2</sup>.
  - d. 1 atm = 101.325 kPa, or 101,325 Pa
6. Dalton's Law of Partial Pressures
  - a. The partial pressures are the pressures that each individual gas exerts in the mixture of gases.
  - b. They are independent of each other.
7. 543 mm
8. Torr units
  - a. 950. torr
  - b. 1.88 torr
  - c.  $3.61 \times 10^7$  torr
  - d.  $5.78 \times 10^9$  torr
9. More pressure units
  - a. 0.164 atm
  - b. 324,000 Pa or  $3.24 \times 10^5$  Pa
  - c. 40.4 mmHg
10. 166.190 torr
11. 699.8 torr
12. Inverse Proportion
13. The molecules are closer together to each other and the container walls, and strike a given unit area of the container of the walls of the container more frequently.
14. Direct proportion
15. Direct proportion
16. Gas particles that are at a higher temperature will have a greater kinetic energy. Therefore at a constant volume for the container, the gas particles will collide with a given unit area of the walls of the container with more force and more frequently resulting in a greater pressure.
17.  $\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$  Or  $\frac{PV}{T} = k$
18. Gases!
  - a. 100.0 mL
  - b. 1.4 mmHg
19. Calculations:
  - a. 93.3 mL

# Holt Chemistry Concept Review Gas Laws Answers

**ML Yell**



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## **Holt Chemistry Concept Review Gas Laws Answers** Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Holt Chemistry Concept Review Gas Laws Answers**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we will delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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web für Kinder und Jugendliche gibt es für die Asthma Therapie ein spezielles Stufenschema das sich vor allem durch eine  
zusätzliche Therapiestufe vom Schema für Erwachsene unterscheidet Ziel jeder Therapie ist es Kindern mit Asthma ein  
normales Leben zu ermöglichen

*asthma beim Kind was Eltern wissen müssen Asthma.de* - Apr 07 2022

web Novartis ist Asthma erblich als häufiger Risikofaktor für Asthma gerade bei Kindern gilt das verstärkte Auftreten in der  
Familie ist ein Elternteil Astmatiker liegt das Risiko bei rund 25 Prozent sind Vater und Mutter betroffen erhöht es sich auf  
mehr als 50 Prozent

asthmatherapie bei Kindern und Jugendlichen mit fixer ICS - Jun 21 2023

web Aug 20 2021 für Jugendliche gilt in Stufe 4 und 5 bei Langzeittherapie mit ICS und Formoterol dass diese Kombination  
dann auch bei Bedarf zum Einsatz kommen sollte für Stufe 1 und 2 reicht die Evidenz aus den großen Erwachsenenlastigen  
Studien aus um auch bei mildem Asthma eine Kombination aus niedrig dosiertem ICS mit Rasch