

Name _____ Date _____ Class _____

CHAPTER

3

STUDY GUIDE FOR CONTENT MASTERY

Matter—Properties and Changes

Section 3.1 Properties of Matter

In your textbook, read about physical properties and chemical properties of matter.

Use each of the terms below just once to complete the passage.

chemical density	mass properties	physical substance
---------------------	--------------------	-----------------------

Matter is anything with (1) mass and volume. A(2) substance is a form of matter with a uniform and unchanging composition.Substances have specific, unchanging (3) properties that can be observed.Substances have both physical and chemical properties. (4) Physical

properties can be observed without changing a substance's chemical composition. Color,

hardness, and (5) density are examples. Other properties cannot be

observed without changing the composition of a substance. These are called

(6) chemical properties. An example is the tendency of iron to form

rust when exposed to air.

Label each property as either physical or chemical.

- physical 7. Chemical formula H_2O
- chemical 8. Irons react vigorously when exposed to moist air
- chemical 9. Remains unchanged when in the presence of nitrogen
- physical 10. Colorless
- physical 11. Solid at normal temperatures and pressures
- chemical 12. Ability to combine with another substance
- physical 13. Melting point
- physical 14. Liquid at normal temperatures and pressures
- physical 15. Boiling point is 100°C
- physical 16. Conducts electricity
- physical 17. Density is $\frac{1}{2} \frac{\text{g}}{\text{cm}^3}$

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CHAPTER

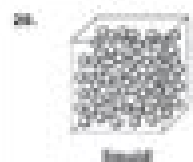
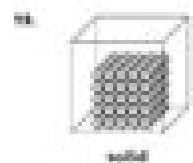
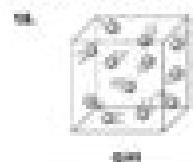
3

STUDY GUIDE FOR CONTENT MASTERY

Section 3.1 continued

In your textbook, read about states of matter.

Label each drawing with one of these words: solid, liquid, gas.



For each statement below, write true or false.

- true 21. All matter that we encounter in everyday life exists in one of these physical forms.
- true 22. A solid has definite shape and volume.
- false 23. A liquid has a definite shape and takes on the volume of its container.
- true 24. A gas has both the shape and the volume of its container.
- false 25. The particles in a gas cannot be compressed into a smaller volume.
- false 26. Liquids tend to contract when heated.
- false 27. The particles in a solid are spaced far apart.
- false 28. The words gas and vapor can be used interchangeably.

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Table of Contents Matter And Change Chapter 8 2 Study Guide Key

1. Understanding the eBook Matter And Change Chapter 8 2 Study Guide Key
 - The Rise of Digital Reading Matter And Change Chapter 8 2 Study Guide Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Matter And Change Chapter 8 2 Study Guide Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Matter And Change Chapter 8 2 Study Guide Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Matter And Change Chapter 8 2 Study Guide Key
 - Personalized Recommendations
 - Matter And Change Chapter 8 2 Study Guide Key User Reviews and Ratings
 - Matter And Change Chapter 8 2 Study Guide Key and Bestseller Lists
5. Accessing Matter And Change Chapter 8 2 Study Guide Key Free and Paid eBooks

- Matter And Change Chapter 8 2 Study Guide Key Public Domain eBooks
 - Matter And Change Chapter 8 2 Study Guide Key eBook Subscription Services
 - Matter And Change Chapter 8 2 Study Guide Key Budget-Friendly Options
6. Navigating Matter And Change Chapter 8 2 Study Guide Key eBook Formats
- ePub, PDF, MOBI, and More
 - Matter And Change Chapter 8 2 Study Guide Key Compatibility with Devices
 - Matter And Change Chapter 8 2 Study Guide Key Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Matter And Change Chapter 8 2 Study Guide Key
 - Highlighting and Note-Taking Matter And Change Chapter 8 2 Study Guide Key
 - Interactive Elements Matter And Change Chapter 8 2 Study Guide Key
8. Staying Engaged with Matter And Change Chapter 8 2 Study Guide Key
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Matter And Change Chapter 8 2 Study Guide Key
9. Balancing eBooks and Physical Books Matter And Change Chapter 8 2 Study Guide Key
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Matter And Change Chapter 8 2 Study Guide Key
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Matter And Change Chapter 8 2 Study Guide Key
- Setting Reading Goals Matter And Change Chapter 8 2 Study Guide Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Matter And Change Chapter 8 2 Study Guide Key
- Fact-Checking eBook Content of Matter And Change Chapter 8 2 Study Guide Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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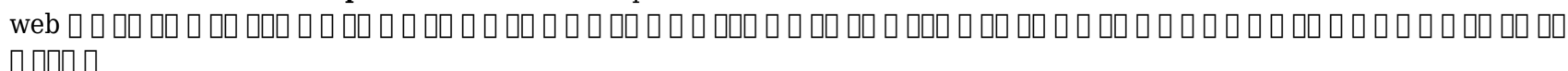
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