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AUSTRALIAN CURRICULUM | FIFTH EDITION





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# **Maths Quest Ac Edition**

Deepti Jain, Nanita Chopra, Kalpana Gairola, S Purkayastha



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buffers typically consist of an acid base pair with the acid and base differing by the presence or absence of a proton a conjugate acid base pair for instance one of the buffers that maintain the ph of human blood involves carbonic acid h 2 co 3 and its conjugate base the bicarbonate ion hco 3

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