

GRAPHING TRENDS OF THE PERIODIC TABLE

CHEMISTRY

NAME: _____

DATE: _____

PERIOD: _____

Graphing Periodic Table Trends

Directions: Using the provided graph paper, graph each of the elemental properties using a line graph. Place the property on the Y-axis and the Atomic number on the X-axis. Draw a dark, solid, vertical line on each graph where each row on the periodic table ends. (i.e. after atomic number 10). Use the graphs to analyze the trends in these properties and to answer the questions on the back.

Atomic Radius (pm)							
Lithium - 3	152	Sodium - 11	190	Potassium - 19	243	Rubidium - 27	265
Beryllium - 4	112	Magnesium - 12	145	Calcium - 20	194	Strontium - 28	219
Boron - 5	87	Aluminum - 13	175	Gallium - 31	136	Indium - 49	156
Carbon - 6	67	Silicon - 14	111	Germanium - 52	125	Tin - 50	140
Nitrogen - 7	56	Phosphorus - 15	98	Arsenic - 33	114	Antimony - 51	133
Oxygen - 8	48	Sulfur - 16	88	Selenium - 34	103	Tellurium - 52	123
Fluorine - 9	42	Chlorine - 17	79	Bromine - 35	94	Iodine - 53	115

Ionization Energy (kJ/mol)							
Lithium - 3	520	Sodium - 11	496	Potassium - 19	419	Rubidium - 27	403
Beryllium - 4	900	Magnesium - 12	738	Calcium - 20	590	Strontium - 28	550

Atomic Radius



Graphing Trends In The Periodic Table Key

Gareth Pritchard



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Table of Contents Graphing Trends In The Periodic Table Key

1. Understanding the eBook Graphing Trends In The Periodic Table Key
 - The Rise of Digital Reading Graphing Trends In The Periodic Table Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Graphing Trends In The Periodic Table Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Graphing Trends In The Periodic Table Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Graphing Trends In The Periodic Table Key
 - Personalized Recommendations
 - Graphing Trends In The Periodic Table Key User Reviews and Ratings
 - Graphing Trends In The Periodic Table Key and Bestseller Lists
5. Accessing Graphing Trends In The Periodic Table Key Free and Paid eBooks

- Graphing Trends In The Periodic Table Key Public Domain eBooks
 - Graphing Trends In The Periodic Table Key eBook Subscription Services
 - Graphing Trends In The Periodic Table Key Budget-Friendly Options
6. Navigating Graphing Trends In The Periodic Table Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Graphing Trends In The Periodic Table Key Compatibility with Devices
 - Graphing Trends In The Periodic Table Key Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Graphing Trends In The Periodic Table Key
 - Highlighting and Note-Taking Graphing Trends In The Periodic Table Key
 - Interactive Elements Graphing Trends In The Periodic Table Key
 8. Staying Engaged with Graphing Trends In The Periodic Table Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Graphing Trends In The Periodic Table Key
 9. Balancing eBooks and Physical Books Graphing Trends In The Periodic Table Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Graphing Trends In The Periodic Table Key
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Graphing Trends In The Periodic Table Key
 - Setting Reading Goals Graphing Trends In The Periodic Table Key
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Graphing Trends In The Periodic Table Key
 - Fact-Checking eBook Content of Graphing Trends In The Periodic Table Key
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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