



## GRADE 10 PHYSICAL SCIENCES

### NOVEMBER EXAMINATION PAPER 2: CHEMISTRY

#### SECTION A

##### QUESTION 1

Various options are provided as possible answers to the following questions. Choose the correct answer and write ONLY the LETTER (A-D) next to the QUESTION/NUMBER in your answer book.

1.1 Which of the following is an example of a mixture?

- A distilled water
- B diamond
- C sodium chloride
- D syngas

1.2 Element with the electron configuration  $1s^2, 2s^2, 2p^4$

- A oxygen
- B carbon
- C helium
- D nitrogen

1.3 The chemical formula of sodium sulfide

- A NaS
- B  $Na_2S$
- C  $NaS_2$
- D  $Na_2S_2$

1.4 Strongest intermolecular force found in water?

- A ion-dipole
- B Dipole-dipole
- C Van der Waals
- D Hydrogen bond

1.5 The particles of a gas ...

- A are tightly packed
- B vibrate against each other
- C move freely and rarely touch
- D move over one another

1.6 Element with the highest electronegativity?

- A Hydrogen
- B Helium
- C Oxygen
- D Fluorine

1.7 When an atom of Group 16 ionises to become an ion, the

- A mass number of the atom will increase
- B atomic number of the atom will decrease
- C number of electrons will decrease
- D number of electrons will increase

1.8 An element with the following Aufbau diagram will be classified as a

|             |    |
|-------------|----|
| $1s^2$      | 2s |
| $2s^2 2p^4$ | 3s |
| $3s^2$      | 3p |
| $3d^5$      | 4s |

- A noble gas
- B halogen
- C alkali earth metal
- D alkali metal

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**M Mosston**



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## **Table of Contents Grade 10 Physical Science Exemplar 2013 November**

1. Understanding the eBook Grade 10 Physical Science Exemplar 2013 November
  - The Rise of Digital Reading Grade 10 Physical Science Exemplar 2013 November
  - Advantages of eBooks Over Traditional Books
2. Identifying Grade 10 Physical Science Exemplar 2013 November
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Grade 10 Physical Science Exemplar 2013 November
  - User-Friendly Interface
4. Exploring eBook Recommendations from Grade 10 Physical Science Exemplar 2013 November
  - Personalized Recommendations
  - Grade 10 Physical Science Exemplar 2013 November User Reviews and Ratings
  - Grade 10 Physical Science Exemplar 2013 November and Bestseller Lists
5. Accessing Grade 10 Physical Science Exemplar 2013 November Free and Paid eBooks
  - Grade 10 Physical Science Exemplar 2013 November Public Domain eBooks
  - Grade 10 Physical Science Exemplar 2013 November eBook Subscription Services
  - Grade 10 Physical Science Exemplar 2013 November Budget-Friendly Options
6. Navigating Grade 10 Physical Science Exemplar 2013 November eBook Formats

- ePub, PDF, MOBI, and More
  - Grade 10 Physical Science Exemplar 2013 November Compatibility with Devices
  - Grade 10 Physical Science Exemplar 2013 November Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Grade 10 Physical Science Exemplar 2013 November
  - Highlighting and Note-Taking Grade 10 Physical Science Exemplar 2013 November
  - Interactive Elements Grade 10 Physical Science Exemplar 2013 November
8. Staying Engaged with Grade 10 Physical Science Exemplar 2013 November
- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Grade 10 Physical Science Exemplar 2013 November
9. Balancing eBooks and Physical Books Grade 10 Physical Science Exemplar 2013 November
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Grade 10 Physical Science Exemplar 2013 November
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Grade 10 Physical Science Exemplar 2013 November
- Setting Reading Goals Grade 10 Physical Science Exemplar 2013 November
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Grade 10 Physical Science Exemplar 2013 November
- Fact-Checking eBook Content of Grade 10 Physical Science Exemplar 2013 November
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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