

Realidades 1

Capítulo 5A

Practice Workbook Answers

SA-1

A.

hermano under her brother
 hermana (twice) once under each
 sister
 primo (twice) once under each
 male cousin
 prima under female cousin
 tío under her aunt
 tío under her uncle
 padre under her father
 madre under her mother
 abuela under her grandmother
 abuelo under her grandfather
 gato under the cat
 perro under the dog

B.

1. no
2. sí
3. sí
4. sí
5. no

SA-2

A.

1. tía
2. hermanos
3. primo
4. abuela
5. padrastro
6. hija
7. prima
8. tíos
9. padres
10. hermanastros

B.

mi madrastra

SA-3

MAMÁ: celebrar / tiene
 TÍA LUCAS: pastel
 MAMÁ: decoraciones
 TÍA LUCAS: Sólo / globos / luces
 MAURO: sacar fotos / regalos
 MAMÁ: la piñata / dulces

SA-4

1. Gabriela es la hermana menor de Raúl.
2. La fiesta es para Gabriela.
3. Es una fiesta de cumpleaños.
4. Raúl decora con los globos y el pastel.
5. El papá tiene los regalos y los globos.

6. Raúl va a hacer un video de la fiesta.
7. Raúl, Gabriela, papá, mamá, los primos, los tíos y los abuelos van a estar en la fiesta.
8. La fiesta va a empezar a las cinco.

SA-5

1. FRANCISCO: _____
 CARMEN: Tengo
 FRANCISCO: Tiene
 CARMEN: Tiene
 FRANCISCO: Siem
 ELISABETH: Siem
 CARMEN: tengo
 ELISABETH: tenemos / tiene / tienes
 CARMEN: tengo
 FRANCISCO: Siem
 JOSÉ: tengo
 FRANCISCO: tengo
 MARIANO: tiene
 RAÚL: _____
 JOSÉ: tenemos
 RAÚL: tengo

SA-6

A.

Row 1: mi hijo, mi tía, mis
 abuelos, _____
 Row 2: tu hijo, _____, tus abuelos,
 tus hermanas
 Row 3: _____, su tía, sus abuelos,
 sus hermanas
 Row 4: nuestro hijo, nuestra tía,
 _____, nuestras hermanas

B.

1. Nuestros
2. tu
3. Mi
4. su
5. Sus
6. mis
7. nuestras
8. su
9. Su
10. Su

SA-7

1. Mis tíos tienen su papel picado.
2. Alicia tiene sus globos.
3. Tú tienes tu piñata.
4. Nosotros tenemos nuestras luces.
5. Yo tengo mi regalo.
6. Ud. tiene su pastel.
7. La profesora Méndez tiene sus flores.
8. Nosotros tenemos nuestras fotos.

Crucigramas (SA-8)

Across:

5. tía
7. tío
9. hermana
11. flor
13. padre
16. abuela
17. hijos
19. gato
20. regalo
21. madre

Down:

1. padrastro
2. hermano
3. pastel
4. picado
6. perro
8. cumpleaños
9. hacer
10. madrastra
12. romper
14. primo
15. saca
18. padres
21. mayor

Organizer (SA-9)

I. Vocabulary: Answers will vary.

II. Grammar

- | | |
|------------|---------------|
| 1. col. 1. | col. 2. |
| tengo | tenemos |
| tiene | tenéis |
| tiene | tienen |
| 2. col. 1. | col. 2. |
| mi / mis | nuestro, -a / |
| | nuestros, -as |
| tu / tus | vuestro, -a / |
| | vuestros, -as |
| su / sus | su / sus |

Navegando 3 Workbook Answers

Rosina Ehmann



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total number of molecules of all the components in the solution

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web solution total mass of the answer 100 g mass of benzene 30 g therefore mass of carbon tet 100 30 g benzene s molar mass C_6H_6 128 g mol⁻¹ hence C_6H_6 s number of moles $\frac{30}{128}$ mol 0 2344 mol molar mass of carbon tet CCl_4 154 g mol⁻¹

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web jan 30 2020 ans the molality of solution is 1 mol kg⁻¹ or 1 m example 02 11 11 g of urea NH_2CONH_2 was dissolved in 100 g of water calculate the molarity and molality of the solution given n 14 h 1 c 12 o 16 given mass of solute urea 11 11 g mass of solvent water 100 g 0 1 kg to find molarity of solution solution

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web solution moles of CH_3OH 5 532 0 17 mole moles of H_2O 40 182 2 moles therefore according to the equation mole fraction of CH_3OH 0 172 2 0 17 mole fraction of CH_3OH 0 073 to solve more examples on mole fraction formula and practice more question please visit byju s com

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web you may work out the mole fractions on your own problem 5 what is the mass of a sample of a 0 449 molal kbr that contains 2 92 kg of water 1 molality moles solute divided by kilograms solute 0 449 mol kg x 2 92 kg x 1 31108 mol of kbr 2 moles times molar mass equals grams 1 31108 mol 119 0023 g mol 156 g kbr

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web calculate the mass percent mole fraction molality and molarity of hexane C_6H_{14} 0 660 g cm⁻³ when it is mixed with heptane C_7H_{16} 0 684 g cm⁻³ in a 35 0 55 0 ml ratio assume that the volumes are additive

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web notice that the mole fraction has no units on it and is written as a decimal value do not change it to percent note of caution you could see the term mole percent it is simply the mole fraction multiplied by 100 for example in the problem just below the mole fraction of cinnamic acid is 0 2885 its mole percent would be 28 85

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web jan 8 2022 mol % x 100 the sum of all of the mole percents of a mixture equals 100 mole fraction properties and advantages the mole fraction offers advantages over some of the other units of concentration unlike molarity mole fraction is *worksheet mole fraction ms fleming* - Jun 11 2023

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web a derivatization that achieves this goes as molarity moles of solute volume of solution in liter moles molarity x volume in liter mass moles x molar mass numerical answers 1

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web since the mole fraction of the o 2 is 0 050 it contributes 0 050 or if you will 5 0 of the total pressure thus total pressure is 0 21 0 050 4 2 atm problem 9 a sample of 1 43 g of helium and an unweighed quantity of o 2 are mixed in a flask at room temperature