

Name: _____ Date: _____ Period: _____

SOLUBILITY WORKSHEET

Use the solubility curves on the attached page to answer the following questions.

1. These solubility curves are based on the solubility (in grams) of various salts dissolving in _____ grams of water. In general for solids, as temperature increases, the solubility of the salt _____.
2. Most and least soluble at different temperatures:
 - a. Which salt is **least** soluble at 20°C? _____
 - b. Which salt is **most** soluble at 10°C? _____
 - c. For which salt does the solubility change the least? _____
3. How many grams of each solute below is needed to saturate 100g of water at 80 °C to prepare the following solutions?
 - a. KNO_3 _____
 - b. KCl _____
 - c. NaCl _____
4. How many grams of each solute below are needed to make a **saturated** solution at 50 °C?
 - a. KClO_3 _____
 - b. NaNO_3 _____
 - c. KNO_3 _____
5. At what temperature are the following solutes equally soluble in 100g of water?
 - a. NaNO_3 and KNO_3 _____
 - b. KNO_3 and NaCl _____
 - c. NaCl and KCl _____
6. The quantity of solute per 100 g of water used in making some solutions is listed below. Is the solution **saturated**, **unsaturated**, or **supersaturated**?

Solute	Temperature	
a. 103 g KNO_3	60 °C	_____
b. 40 g KCl	40 °C	_____

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Zhe Xu



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