

Name : _____ Date : _____



Gram Formula Mass Worksheet



1. What is the percentage by mass composition of Iron (III) oxide?
2. Calculate the molar mass of $C_3H_5N_3O_9$ (Nitroglycerin, an explosive)
3. How many atoms are found in 1.55 grams of chlorine gas?
4. When silver sold for \$16.00 per ounce, how many silver atoms could you buy for 10.00 dollars?
5. How many grams of carbon are there in 14.0 g of $Pb(C_2H_5)_4$ (tetraethyllead, a gasoline additive)?
6. A mixture contains 10.00 g of NaBr and 5.00 g of BaBr₂. What is the total number of moles of bromide ions in the mixture?
7. Determine the moles of sodium in 7.22×10^{100} kg of $Na_2S_2O_3$
8. How many atoms of Zn would contain the same number of grams as 7.54×10^{-6} mg of Cu?

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