

III. CHEMISTRY NETWORK LENSES

CHEMICAL REACTIONS

GRADE 10-11

1. Hydrogen gas is produced with large quantities of carbon monoxide. The hydrogen then undergoes a series of chemical reactions (process 1 below) to produce ethylene, which is then polymerized (process 2 below) to produce polyethylene.

	Process 1: Hydrogenation	Process 2: Polymerization
A.	Hydrogen	Hydrogen
B.	Carbon monoxide	Carbon monoxide
C.	Ethylene	Ethylene
D.	Polyethylene	Polyethylene

2. Draw arrows in each of the empty boxes below showing the following reactions:

Hydrogen Carbon monoxide

The polyethylene is produced from ethylene:

Hydrogen Carbon monoxide
Polyethylene

Hydrogen Carbon monoxide
Polyethylene

Write in the boxes with species from Process 1

	Hydrogen	Carbon monoxide
A.	Hydrogen	Carbon monoxide
B.	Carbon monoxide	Carbon monoxide
C.	Ethylene	Ethylene
D.	Polyethylene	Polyethylene

3. The polyethylene is produced from ethylene. The ethylene is produced from ethylene. Write in the boxes

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Table of Contents Kinetics Paper One Ib Questions

1. Understanding the eBook Kinetics Paper One Ib Questions
 - The Rise of Digital Reading Kinetics Paper One Ib Questions
 - Advantages of eBooks Over Traditional Books
2. Identifying Kinetics Paper One Ib Questions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Kinetics Paper One Ib Questions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Kinetics Paper One Ib Questions
 - Personalized Recommendations
 - Kinetics Paper One Ib Questions User Reviews and Ratings

- Kinetics Paper One Ib Questions and Bestseller Lists
- 5. Accessing Kinetics Paper One Ib Questions Free and Paid eBooks
 - Kinetics Paper One Ib Questions Public Domain eBooks
 - Kinetics Paper One Ib Questions eBook Subscription Services
 - Kinetics Paper One Ib Questions Budget-Friendly Options
- 6. Navigating Kinetics Paper One Ib Questions eBook Formats
 - ePub, PDF, MOBI, and More
 - Kinetics Paper One Ib Questions Compatibility with Devices
 - Kinetics Paper One Ib Questions Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Kinetics Paper One Ib Questions
 - Highlighting and Note-Taking Kinetics Paper One Ib Questions
 - Interactive Elements Kinetics Paper One Ib Questions
- 8. Staying Engaged with Kinetics Paper One Ib Questions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Kinetics Paper One Ib Questions
- 9. Balancing eBooks and Physical Books Kinetics Paper One Ib Questions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Kinetics Paper One Ib Questions
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Kinetics Paper One Ib Questions
 - Setting Reading Goals Kinetics Paper One Ib Questions
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Kinetics Paper One Ib Questions
 - Fact-Checking eBook Content of Kinetics Paper One Ib Questions
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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