Physics DF025

Learning Outcome:

- 14.2 Kinetic theory of gases (1 hour)
- At the end of this chapter, students should be able to:
- State the assumptions of kinetic theory of gases.
- Apply the equations of ideal gas,

$$PV = \frac{1}{3} Nm < v^2 >$$

and pressure,

$$P = \frac{1}{3} \rho < v^2 >$$

in related problems.

 Explain and use root mean square (rms) speed,

$$\langle v^2 \rangle = 3 \frac{kT}{m}$$

of gas molecules.

Kinetic Theory Chapter 14 Section Reinforcement

Clifford Lane

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