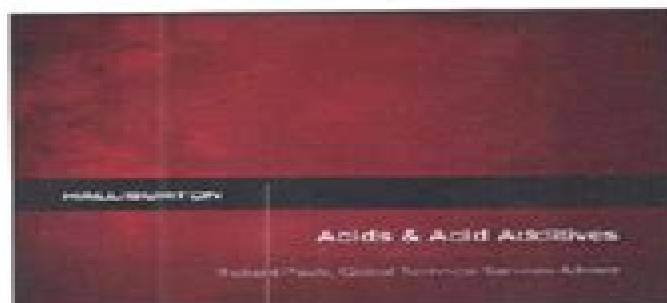


Types of Acid

- **Mineral Acids** *(strong)*
 - HCl
 - HF
- **Organic Acids** *(weak) acts like True Buffers*
 - Acetic
 - Formic
 - Citric
 - Others used for specific purposes, such as in HF acid systems

Types of Acid – Mineral Acids - HCl *(strong)*

- **Why Used**
 - Used to stimulate carbonate formations
 - Used in combination with HF acid to remove damage from sandstone formations
 - High concentrations can be used
 - Relatively inexpensive
 - Reaction byproducts do not precipitate
 - Can be inhibited on most metal types over most conditions

Types of Acid – Mineral Acids - HCl

- **Possible Problems**
 - Difficult to inhibit at high temperatures
 - Some minerals are sensitive to HCl at elevated temperatures
 - May cause HCl or HCl / Iron induced sludging in some crude oils
 - Limited availability in some parts of the world

Types of Acid – Mineral Acids - HCl

Chemical Reaction of HCl with Carbonate Minerals

$$2\text{HCl} + \text{CaCO}_3 \rightarrow \text{CO}_2\uparrow + \text{CaCl}_2 + \text{H}_2\text{O}$$
$$4\text{HCl} + \text{CaMg}(\text{CO}_3)_2 \rightarrow 2\text{CO}_2\uparrow + \text{MgCl}_2 + \text{CaCl}_2 + 2\text{H}_2\text{O}$$

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