

## Stoichiometry Worksheet 2: Percent Yield

For each of the problems below:

- Write the balanced chemical equation
- Identify the given (with units) and what you want to find (with units)
- Show set up with units. Check sig figs, give final answer with units and label.

1. Using the Hoffman apparatus for electrolysis, a chemist decomposes 36 g of water into its gaseous elements. How many grams of hydrogen gas should she get (theoretical yield)?



Before  $\underline{2 \text{ mol}}$   $\underline{0 \text{ mol}}$   $\underline{0 \text{ mol}}$

Change  $\underline{-2 \text{ mol}}$   $\underline{+2 \text{ mol}}$  \_\_\_\_\_

After  $\underline{0 \text{ mol}}$   $\underline{2 \text{ mol}}$

$$36 \text{ g H}_2\text{O} \times \frac{1 \text{ mol H}_2\text{O}}{18.0 \text{ g H}_2\text{O}} = 2.0 \text{ mol H}_2\text{O} \times \frac{2 \text{ mol H}_2}{2 \text{ mol H}_2\text{O}} = 2 \text{ mol H}_2 \times \frac{2.0 \text{ g H}_2}{1 \text{ mol H}_2} = 4.0 \text{ g H}_2$$

2. Recall that liquid sodium reacts with chlorine gas to produce sodium chloride. You want to produce 581 g of sodium chloride. How many grams of sodium are needed?



Before  $\underline{9.94 \text{ mol}}$   $\underline{\text{xs mol}}$   $\underline{0 \text{ mol}}$

Change  $\underline{-9.94 \text{ mol}}$  \_\_\_\_\_  $\underline{+9.94 \text{ mol}}$

After  $\underline{0 \text{ mol}}$   $\underline{9.94 \text{ mol}}$

$$581 \text{ g NaCl} \times \frac{1 \text{ mol NaCl}}{58.5 \text{ g NaCl}} = 9.93 \text{ mol NaCl} \times \frac{2 \text{ mol Na}}{2 \text{ mol NaCl}} = 9.93 \text{ mol Na} \times \frac{23.0 \text{ g Na}}{1 \text{ mol Na}} = 228 \text{ g Na}$$

3. You eat 180.0 g of glucose (90 M&Ms). If glucose,  $\text{C}_6\text{H}_{12}\text{O}_6$ , reacts with oxygen gas to produce carbon dioxide and water, how many grams of oxygen will you have to breathe in to burn the glucose?



Before  $\underline{1.0 \text{ mol}}$   $\underline{\text{xs mol}}$   $\underline{0 \text{ mol}}$   $\underline{0 \text{ mol}}$

Change  $\underline{-1.0 \text{ mol}}$  \_\_\_\_\_  $\underline{-6.0 \text{ mol}}$

After  $\underline{0 \text{ mol}}$   $\underline{\text{xs mol}}$

$$180.0 \text{ g C}_6\text{H}_{12}\text{O}_6 \times \frac{1 \text{ mol C}_6\text{H}_{12}\text{O}_6}{180 \text{ g C}_6\text{H}_{12}\text{O}_6} = 1.00 \text{ mol C}_6\text{H}_{12}\text{O}_6 \times \frac{6 \text{ mol O}_2}{1 \text{ mol C}_6\text{H}_{12}\text{O}_6} = 6.00 \text{ mol O}_2$$

$$6.0 \text{ mol O}_2 \times \frac{32.0 \text{ g O}_2}{1 \text{ mol O}_2} = 192.0 \text{ g O}_2$$

# Laying The Foundation In Chemistry Stoichiometry Answers

**Daniel Barker Murphy, Viateur  
Rousseau**



## **Laying The Foundation In Chemistry Stoichiometry Answers:**

**AP Chemistry: Achieving Excellence** Pasquale De Marco, 2025-07-15 Embark on a transformative learning journey with AP Chemistry Achieving Excellence a comprehensive guidebook designed to empower students in their pursuit of mastering AP Chemistry This invaluable resource provides an in depth exploration of fundamental concepts equipping students with a comprehensive understanding of chemical reactions structures thermodynamics and electrochemistry With clarity and simplicity at its core AP Chemistry Achieving Excellence unravels complex topics making them accessible to students of all learning styles Engaging activities thought provoking exercises and real world examples bring abstract concepts to life fostering a deeper comprehension and retention of knowledge This comprehensive book is more than just a textbook it is an indispensable companion for students aiming for success in AP Chemistry Its user friendly approach abundance of practice opportunities and extensive explanations make it an ideal choice for students seeking mastery in the subject Aspiring chemists and students seeking a solid foundation in chemistry will find AP Chemistry Achieving Excellence an invaluable guide Its diverse learning tools interactive exercises and user friendly explanations make it the perfect resource for achieving academic excellence in AP Chemistry and beyond Unlock the secrets of chemistry and embark on a journey of scientific discovery with AP Chemistry Achieving Excellence Let this book be your trusted companion as you navigate the fascinating world of chemical reactions structures thermodynamics and electrochemistry With AP Chemistry Achieving Excellence by your side you will gain the confidence and skills necessary to tackle the challenging AP Chemistry exam and excel in your future endeavors in chemistry and related fields If you like this book write a review

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hydrogen technology or gain insight into its role in the global energy transition this book offers a thorough and accessible guide to hydrogen s revolutionary potential Explore the remarkable journey of hydrogen and its power to fuel the future of energy     *Chemistry* Steven S. Zumdahl,1989     *The Merck Report* ,1908     **Problem Solving in General Chemistry** Ronald A. DeLorenzo,1981     **Peroxynitrite Detection in Biological Media** Serban Petcu,Sabine Szunerits,Mekki Bayachou,2015-10-16 Peroxynitrite is a powerful oxidiser which can damage a wide array of molecules within cells including DNA and proteins leading to apoptosis inflammation or cancer Peroxynitrite detection and quantification provides critical information in understanding its biological implications Attempts to investigate the behavior of peroxynitrite in vivo and in vitro have been hampered by the difficulty in detecting this highly reactive oxygen species This book presents the current state of the art in this research field with contributions from scientific leaders in the field The chapters make clear the associated challenges and development for selective and sensitive detection of peroxynitrite This book is a timely addition to the literature as the first in the field dedicated to detecting this molecule in vivo It will be welcomed by the community particularly medicinal and analytical chemists developers of sensors and probes and analytical equipment manufacturers

**Foundations of College Chemistry** Daniel Barker Murphy,Viateur Rousseau,1980     *Catalogue* Albion College,1916  
*Abstracts of Papers* American Chemical Society. Meeting,1979     *Directory of Distance Learning Opportunities* Modoc Press, Inc.,2003-02-28 This book provides an overview of current K 12 courses and programs offered in the United States as correspondence study or via such electronic delivery systems as satellite cable or the Internet The Directory includes over 6 000 courses offered by 154 institutions or distance learning consortium members Following an introduction that describes existing practices and delivery methods the Directory offers three indexes Subject Index of Courses Offered by Level Course Level Index Geographic Index All information was supplied by the institutions Entries include current contact information a description of the institution and the courses offered grade level and admission information tuition and fee information enrollment periods delivery information equipment requirements credit and grading information library services and accreditation     **AAAS Science Book List, 1978-1986** Kathryn Wolff,Susan M. O'Connell,Valerie J. Montenegro,1986 A selected and annotated list of science and mathematics books which supplements the AAAS science book list 3rd ed 1970 and the AAAS science book list supplement 1978     **Macmillan Encyclopedia of Chemistry** J. J. Lagowski,1997

Comprehensive and up to date this unique four volume set offers readers a complete overview of the broad spectrum of general chemistry It enables them to obtain a basic yet thorough understanding of matter the processes it undergoes the principles that govern it and the international cast of men and women who have been critical in the development of the science of chemistry From elements atoms and molecules to terochemistry spectroscopy and chemical bonding its clear and concise explanations provide an illuminating and readily comprehensible introduction Key presentations include forty element definition articles each providing basic periodic table information and general information on the element in

question Ninety five biographical articles deal with prominent chemists while other articles provide additional historical context particularly with respect to eighteenth nineteenth and twentieth century developments

**Hawley's Condensed Chemical Dictionary** Gessner Goodrich Hawley, Richard J. Lewis (Sr.), 1993 Hawley's Condensed Chemical Dictionary 15th Edition is a compilation of technical data and descriptive information covering thousands of chemicals and chemical phenomena trade name products processes reactions products and related terminology Updates and expands the coverage from the previous edition Adds entries for notable chemists and Nobel Prize winners equipment and devices natural forms and minerals named reactions and chemical processes Provides concise condensed and prompt definitions of terms and phenomena in chemistry biology biochemistry and more Is ideal for those with only minutes to devote to any given chemical substance or topic

Words on Cassette ,1995

*Wilhelm Ostwald* Robert Smail Jack, Fritz Scholz, 2017-01-25 This book is the translated and commented autobiography of Wilhelm Ostwald 1853 1932 who won the Nobel Prize for Chemistry in 1909 It is the first translation of the German original version *Lebenslinien Eine Selbstbiographie* published by Ostwald in 1926 27 and has been painstakingly translated The book includes comments and explanations helping readers to understand Ostwald's text in the historical context of Germany at the beginning of the 20th century In his autobiography Ostwald describes his impressive research career and his life from his own personal view Readers will find information on how Ostwald immortalized himself through his research on catalysis chemical equilibria technical chemistry and especially as one of the founders of modern physical chemistry His broad interests in science ranging from philosophy to the theory of colors and the idea of a universal scientific language are further remarkable aspects covered This work will appeal to a broad audience of contemporary scientists Wilhelm Ostwald has been tremendously influential for the development of chemistry and science and many of today's best known international scientific schools can be traced back to Ostwald's students Ostwald was active in Germany and what is now Latvia and Estonia while also travelling to the USA England and France In his discussions and analyses of the working conditions of the time readers will find many issues reflected that continue to be of relevance today

Journal of the American Chemical Society American Chemical Society, 1944

*Chemical Engineering Catalog* ,1925

**The Macmillan Guide to Correspondence Study** Modoc Press, 1996

Ceramic Abstracts American Ceramic Society, 1995

Chem 5e Irm Steven S. Zumdahl, 2000-02

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