



Practice Worksheet

1. Rank the following acids from **strongest** to **weakest** acid.

Acid	Structure	Estimated K_a	Estimated pK_a
1	HNO_3		
2	H_2SO_4		
3	HClO_4		
4	H_2SO_3		

- A. $1 > 2 > 3 > 4$
 B. $2 > 3 > 1 > 4$
 C. $1 > 3 > 2 > 4$
 D. $1 > 2 > 4 > 3$

2. Rank the acids according to increasing acidity. Rank all 4 from **strongest** to **weakest**. Be consistent in ranking. Assign each species a number representing its rank.

- A. HNO_2 (acid) HNO_3 (strongest) HNO (weakest) HNO_2 (strongest)
 B. HNO_2 (acid) HNO_3 (strongest) HNO (weakest) HNO_2 (strongest)
 C. HNO_2 (acid) HNO_3 (strongest) HNO (weakest) HNO_2 (strongest)
 D. HNO_2 (acid) HNO_3 (strongest) HNO (weakest) HNO_2 (strongest)

3. Which of the following strong acids is **strongest** in water? Circle the correct choice. Be consistent in ranking.



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Table of Contents June 2012 6ch01

1. Understanding the eBook June 2012 6ch01
 - The Rise of Digital Reading June 2012 6ch01
 - Advantages of eBooks Over Traditional Books
2. Identifying June 2012 6ch01
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an June 2012 6ch01
 - User-Friendly Interface
4. Exploring eBook Recommendations from June 2012 6ch01
 - Personalized Recommendations
 - June 2012 6ch01 User Reviews and Ratings
 - June 2012 6ch01 and Bestseller Lists
5. Accessing June 2012 6ch01 Free and Paid eBooks
 - June 2012 6ch01 Public Domain eBooks

- June 2012 6ch01 eBook Subscription Services
- June 2012 6ch01 Budget-Friendly Options
- 6. Navigating June 2012 6ch01 eBook Formats
 - ePub, PDF, MOBI, and More
 - June 2012 6ch01 Compatibility with Devices
 - June 2012 6ch01 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of June 2012 6ch01
 - Highlighting and Note-Taking June 2012 6ch01
 - Interactive Elements June 2012 6ch01
- 8. Staying Engaged with June 2012 6ch01
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers June 2012 6ch01
- 9. Balancing eBooks and Physical Books June 2012 6ch01
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection June 2012 6ch01
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine June 2012 6ch01
 - Setting Reading Goals June 2012 6ch01
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of June 2012 6ch01
 - Fact-Checking eBook Content of June 2012 6ch01
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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