

**Section 1.1** *continued*

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

19
39
55
87
107
117

atomic mass	19.
atomic number	39.
electron configuration	55.
chemical name	87.
chemical symbol	117.

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 116. It exists in the periodic table.
- _____ 118. It has no bar in periodic table.
- _____ 120. Group 1 elements.
- _____ 122. Elements that are shiny and conduct electricity.
- _____ 124. Group 17 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write *yes* if the statement is true, if the statement is false, change the underlined word or phrase to make it true.

- _____ 126. There are two representative elements of elements.
- _____ 128. Most of the elements in the periodic table are metals.
- _____ 130. Group 1A elements except for hydrogen are known as the alkali metals.
- _____ 132. Group 2A elements are the alkaline earth metals.
- _____ 134. Group 7A elements are highly reactive nonmetals known as halogens.
- _____ 136. Group 8A elements are very reactive elements known as noble gases.
- _____ 138. Elements have properties of both metals and nonmetals metals.

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